

Supporting information for: *Cyclodextrin Ion Channels*

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Voltage Clamp Data Acquisition

A model BC-525A bilayer clamp (Warner Instrument Corp.) was used for planar bilayer experiments. The analogue output was filtered with an 8-pole Bessel filter (Frequency Devices, model 902) and digitized with a 330 kHz digitizer (Axon Instruments, Digidata 1200A). Data acquisition was controlled by the pClamp8 software package (Axon Instruments). Data were collected at 10 Hz, analogue filtered at 1 Hz, and digitally filtered at 50 Hz. The headstage and the bilayer chamber (3 mL polystyrene cuvette with 250 μ m diameter aperture held in a 5 mL PVC holder) were placed on a floating table and electrically shielded by a grounded aluminum Faraday cage. Agar salt bridges (2 M KNO₃ in 1% Agar) were used to stabilize junction potentials and were employed between the electrolyte in each well of the cell and Ag/AgCl electrodes. Electrolyte solutions were prepared from high purity salts and nanopure water.

A stock solution of diphytanoyl phosphatidylcholine (diPhyPC) in chloroform (Avanti Polar Lipids; shipped on dry ice) was divided into sealed glass vials under an argon atmosphere and stored at -12 °C. For use in an experiment, a stream of dry nitrogen was passed through the vial for 1 hour. The dried lipid was diluted with decane to give a solution concentration of 25 mg/mL in lipid.

Bilayers were formed by either brushing or dipping: after lipid in decane had been introduced by brushing, a lipid/ decane film formed on the surface of the electrolyte, and bilayers could then be formed by withdrawal of 2-3 mL of electrolyte from the cell holder by syringe to expose one face of the aperture to the air-water interface held in the cell holder, followed by reintroduction of the electrolyte to oppose monolayers across the aperture in the cuvette. Bilayer quality was monitored via the capacitance and stability under applied potential, using the criteria previously described¹. The measured voltage was applied with respect to the *trans* (cuvette) side of the bilayer, making the *trans* side the relative ground. Digitized data files were analyzed using the pClamp10 suite of programs.

The compounds are introduced to the membrane in two ways, depending on the solvent in which the compound can be dissolved:

Direct injection - all injection experiments utilized bilayers that were apparently stable at 100 mV for periods of 20 minutes or more. Aliquots (1-5 μ L of transporter solutions in MeOH) were injected with a microliter syringe as close as possible to the bilayer in the free well of the cuvette holder (*cis* side), and gently stirred with a stream of nitrogen for 5 minutes.

Pre-mixed into lipid - in this method, 1mol% of compound (in CDCl₃ or MeOH-d₄) was added to the diPhyPC/CHCl₃ solution, and solvent removed with a stream of N₂, and bilayer membrane prepared by brushing/dipping as described above. Most of the bilayers formed with this method gave bilayers with good quality.

Of the two methods, direct injection is preferable, as it allows monitoring of pristine bilayer prior to compound introduction. Following direct injection, channel behaviour typically appears within 20

minutes of compound introduction, and persists over period of hours. Once stabilized, continuous data acquisition of at least 30-60 minutes is required to provide sufficient statistical power for the power-law analysis; shorter acquisition periods (10 minutes) are sufficient to characterize other types of behaviors.

Power Law Fitting Procedure

Fitting experimental data to a power law requires two distinct steps. The first step transforms the irregular current trace into a list of opening times; this list is then fitted to a power law distribution.

Event List Generation Manipulation of the digitally filtered traces was carried out using Clampfit 10 of the pClamp suite. A customized threshold search was used to generate the list of events. The threshold was set across the fluctuating section of the trace to maximize the number of events. Within that segment, α is insensitive to the choice of threshold. A minimum duration was fixed at 50ms. The threshold search automatically logs *event start* and *event end* from which the duration can be calculated. The resultant values were exported to the fitting program.

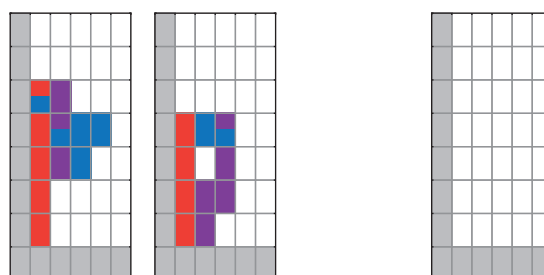
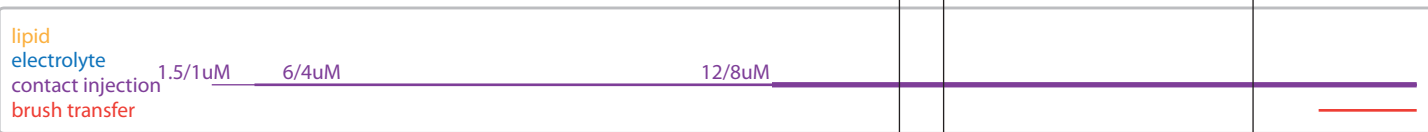
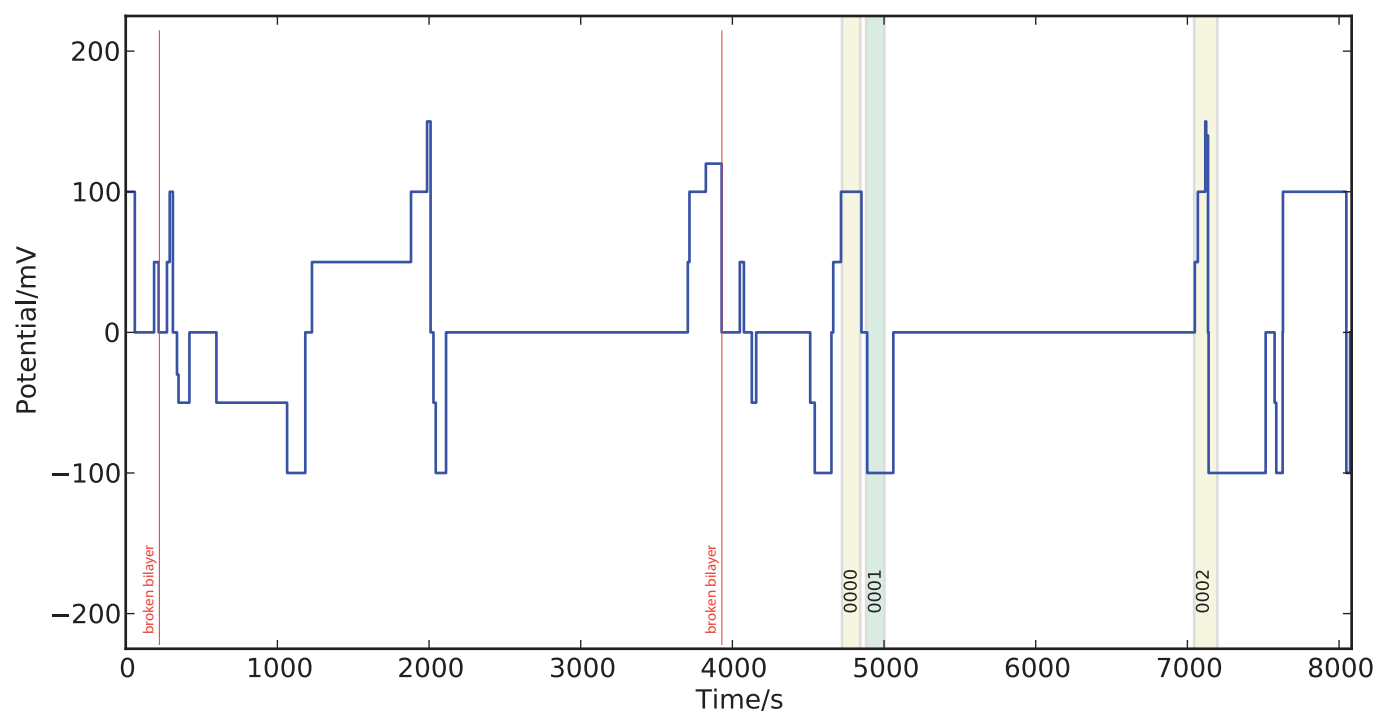
Power Law Fitting The list of opening durations, obtained above as a plain-text file, can then be fitted using the method of Clauset *et al*², implemented in python³. The code performs the Maximum Likelihood Estimate fit, and provides α , $x_{\min}$, n , and p-value as outputs.

Summary of bilayer activity

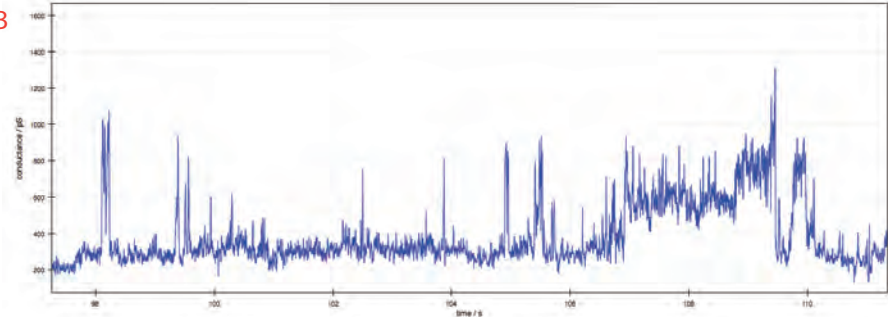
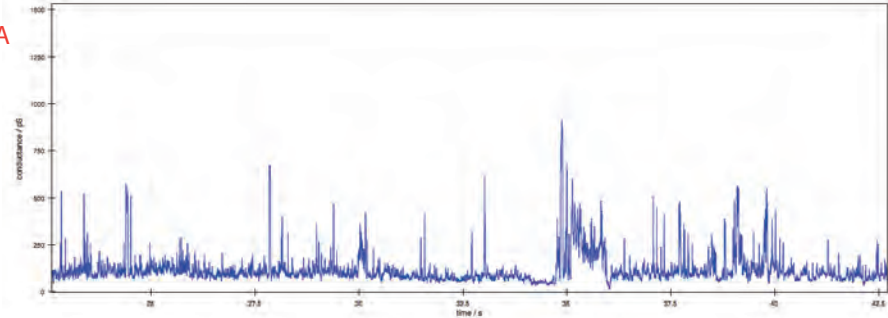
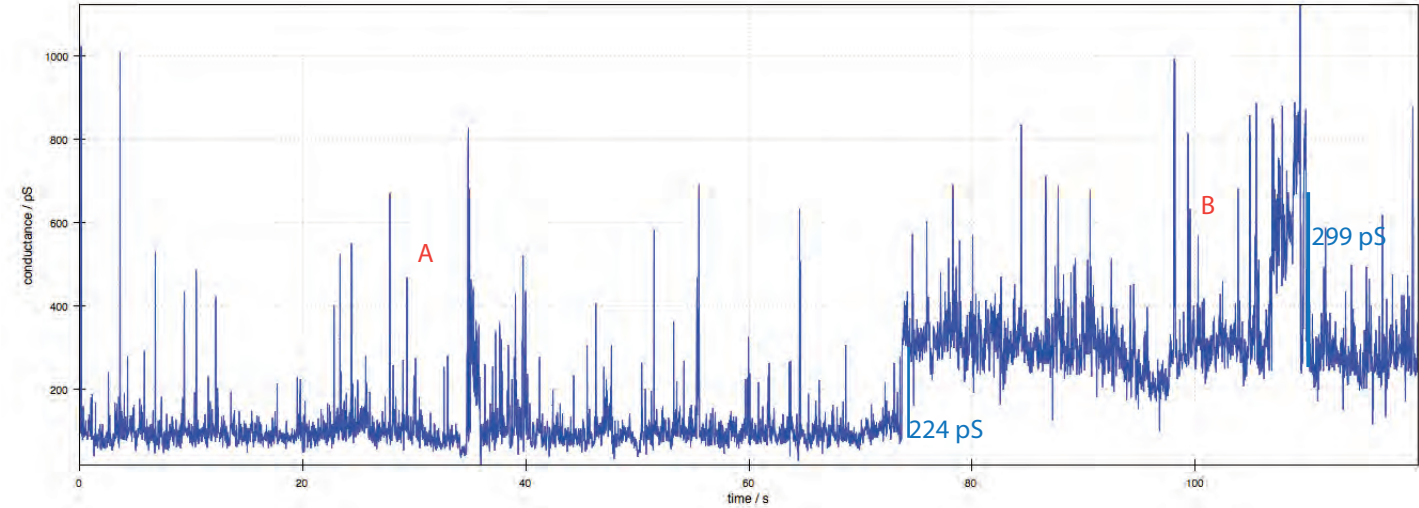
Annotated activity grids, as well as full conductance records (and expansions where appropriate), are provided below for every compound studied. The activity grids were prepared as previously described⁴. The summaries are arranged first by compound, then individual experiments. Within each experiment, the first page(s) summarizes the experimental conditions as well as activity grids charted; subsequent pages shows the full conductance record as the top panel, with expansions indicated by corresponding letters.

1. Fyles, T. M.; Knoy, R.; Müllen, K.; Sieffert, M., Membrane activity of isophthalic acid derivatives: ion channel formation by a low molecular weight compound. *Langmuir* **2001**, *17*, 6669-6674.
2. Clauset, A.; Shalizi, C. R.; Newman, M. E. J., Power-law distributions in empirical data. *SIAM Rev.* **2009**, *51*, 661-703.
3. Ginsberg, A. <<http://code.google.com/p/aqpy/wiki/PowerLaw>> (accessed November 7).
4. (a) Chui, J. K. W. A New Paradigm for Voltage-Clamp Studies of Synthetic Ion Channels. University of Victoria, Victoria, 2011; (b) Chui, J. K. W.; Fyles, T. M.; Luong, H., Planar bilayer activities of linear oligoester bolaamphiphiles. *Beilstein J. Org. Chem.* **2011**, *7*, 1562-1569.

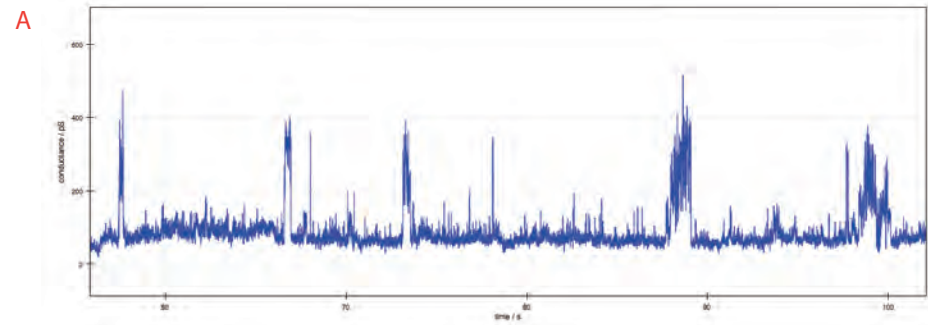
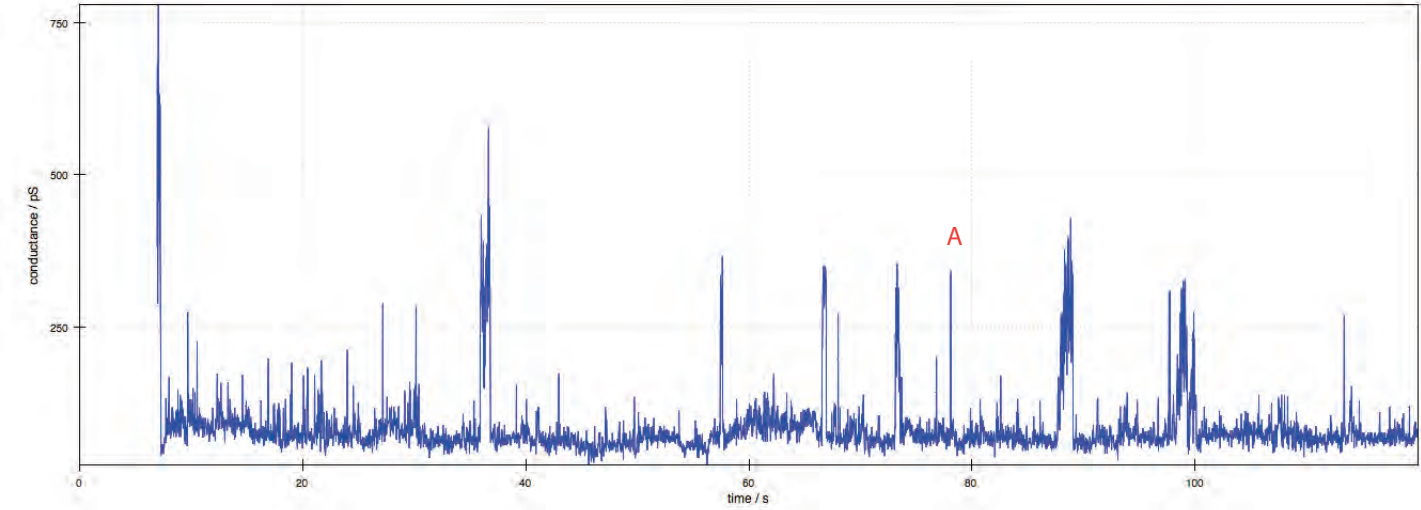
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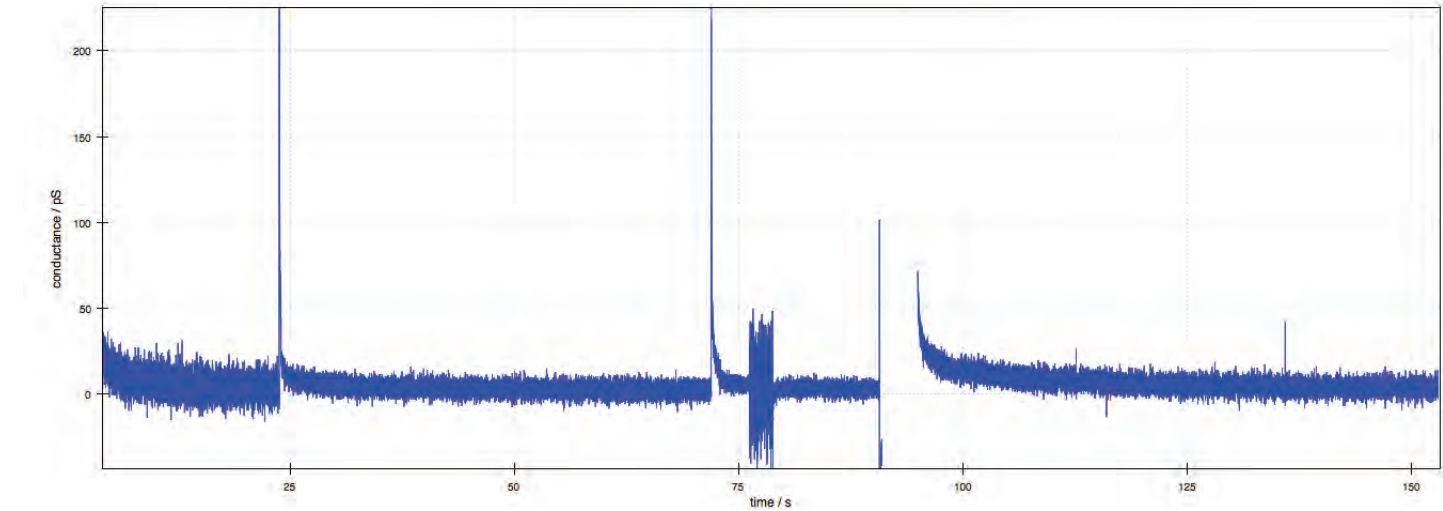
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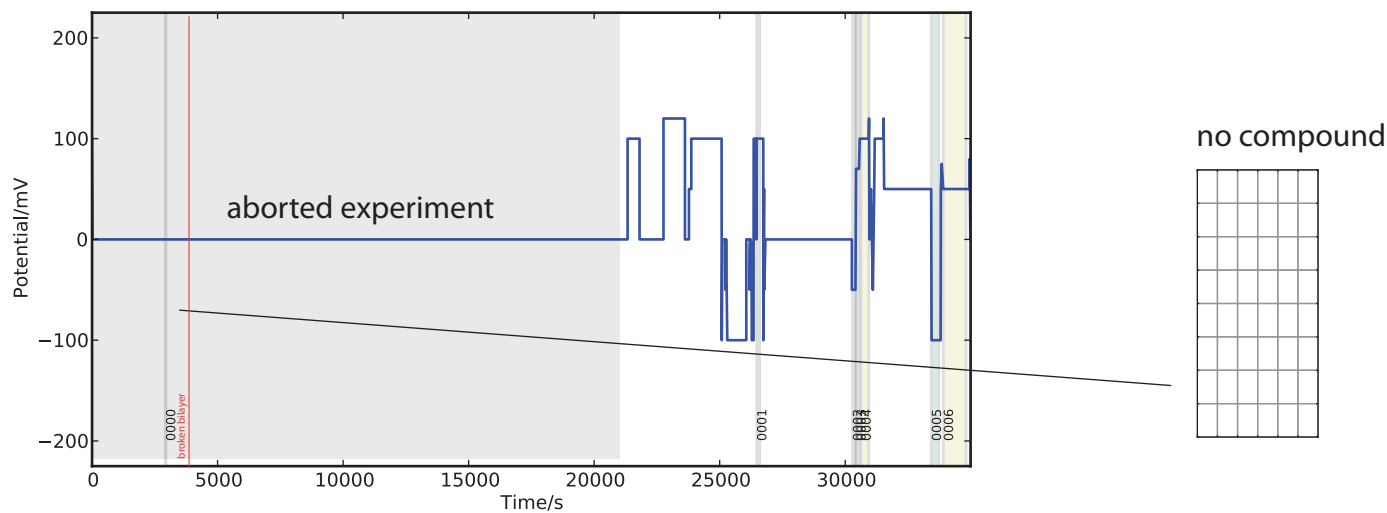
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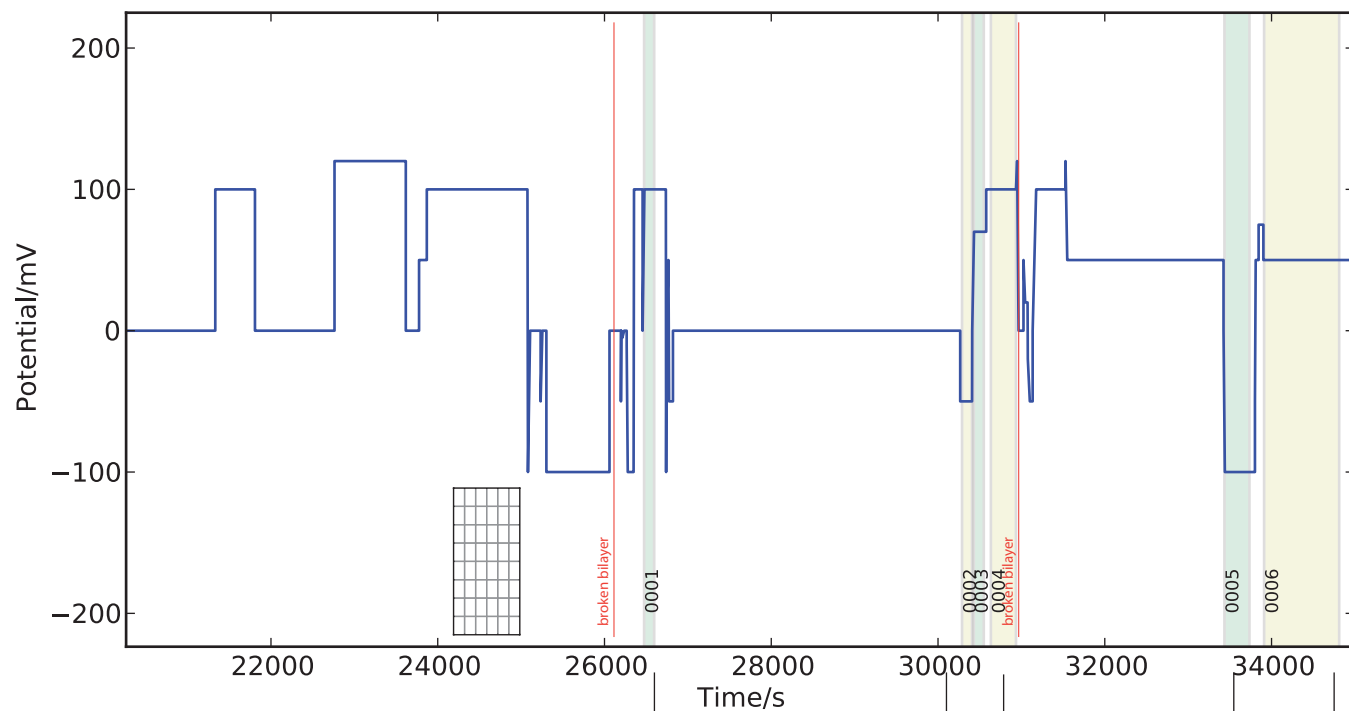
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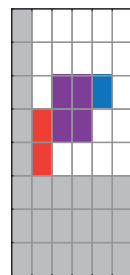
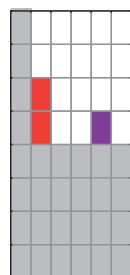
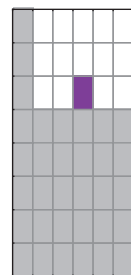
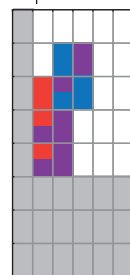
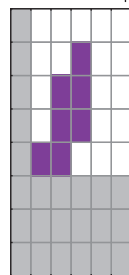
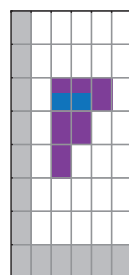


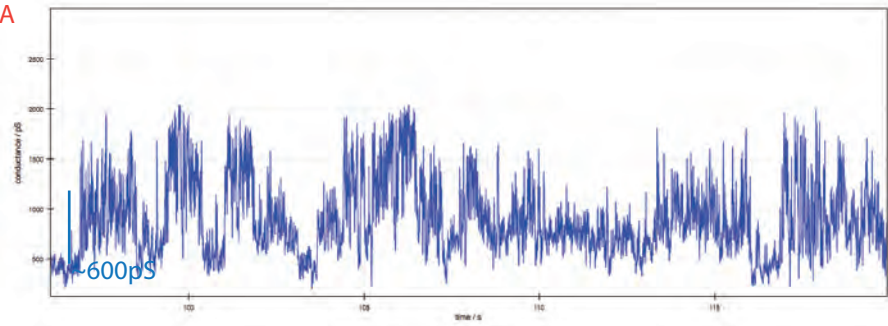
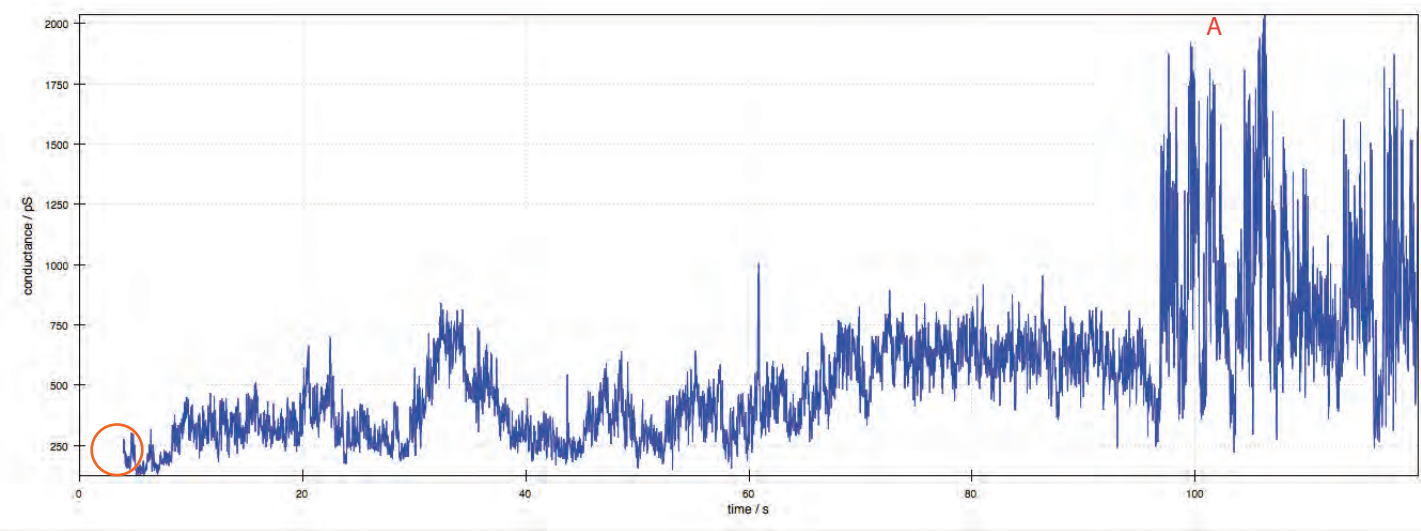
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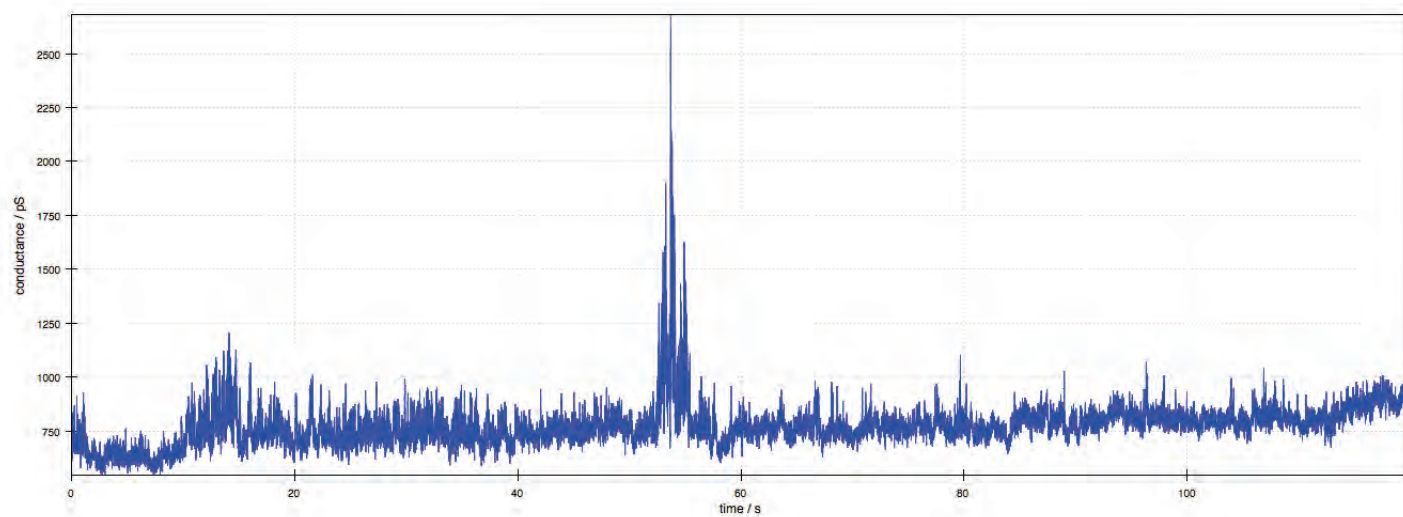
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electrolyte 11/7uM
contact injection
brush transfer

19/12uM total

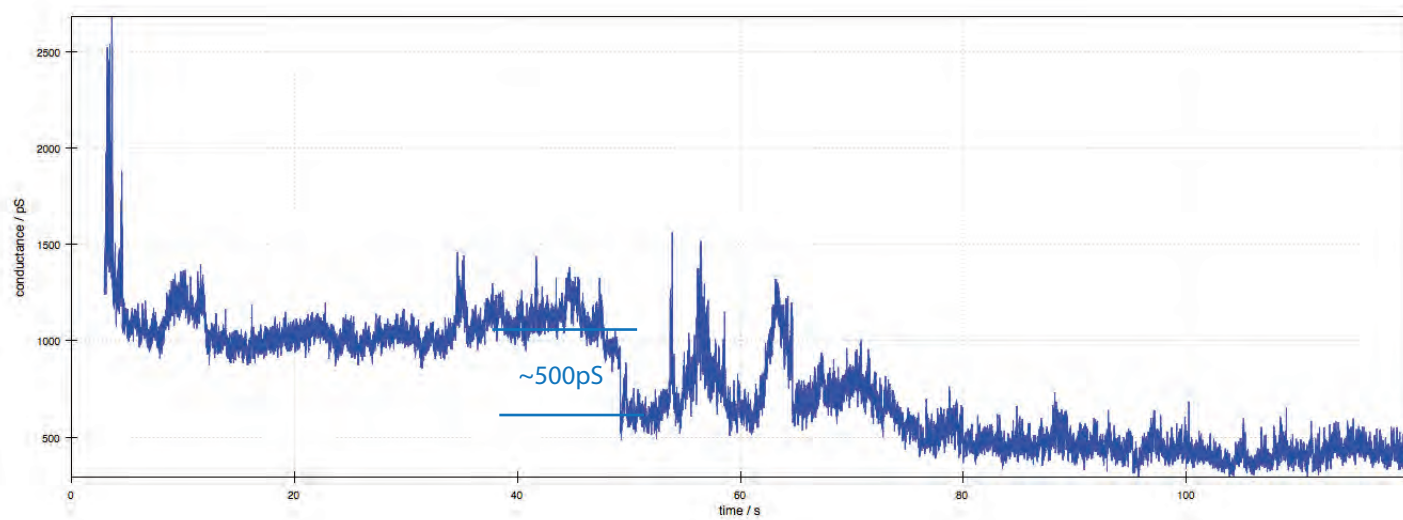


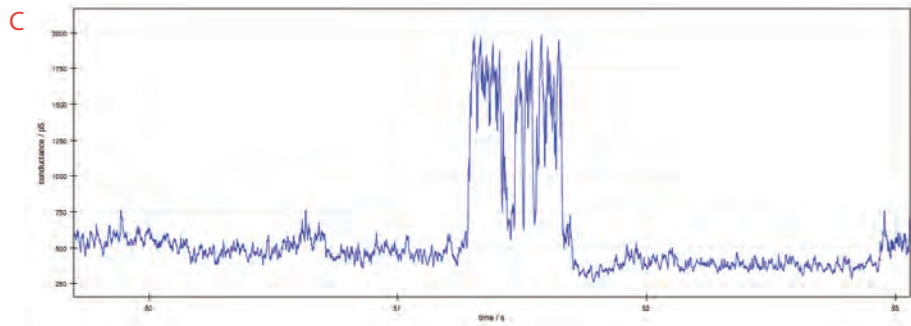
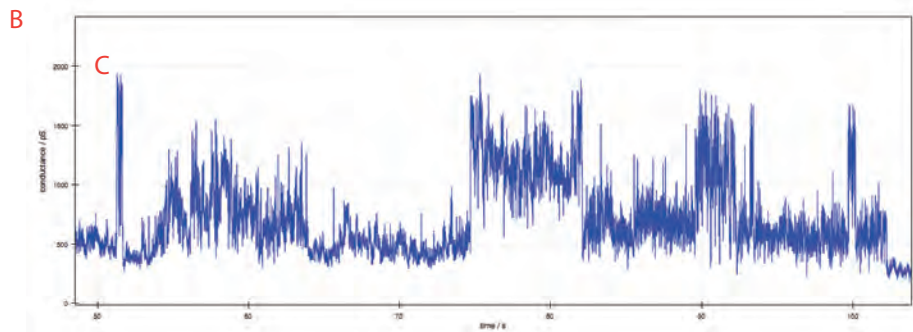
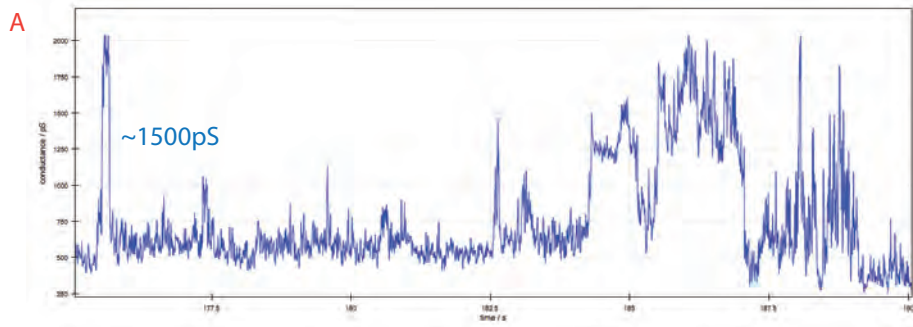
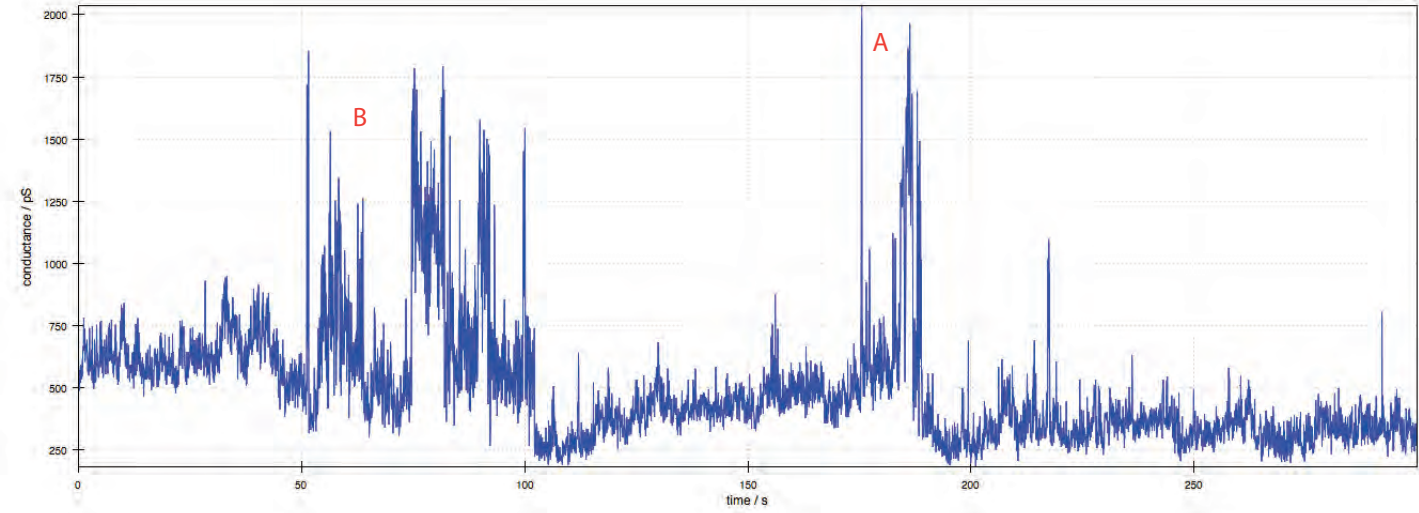


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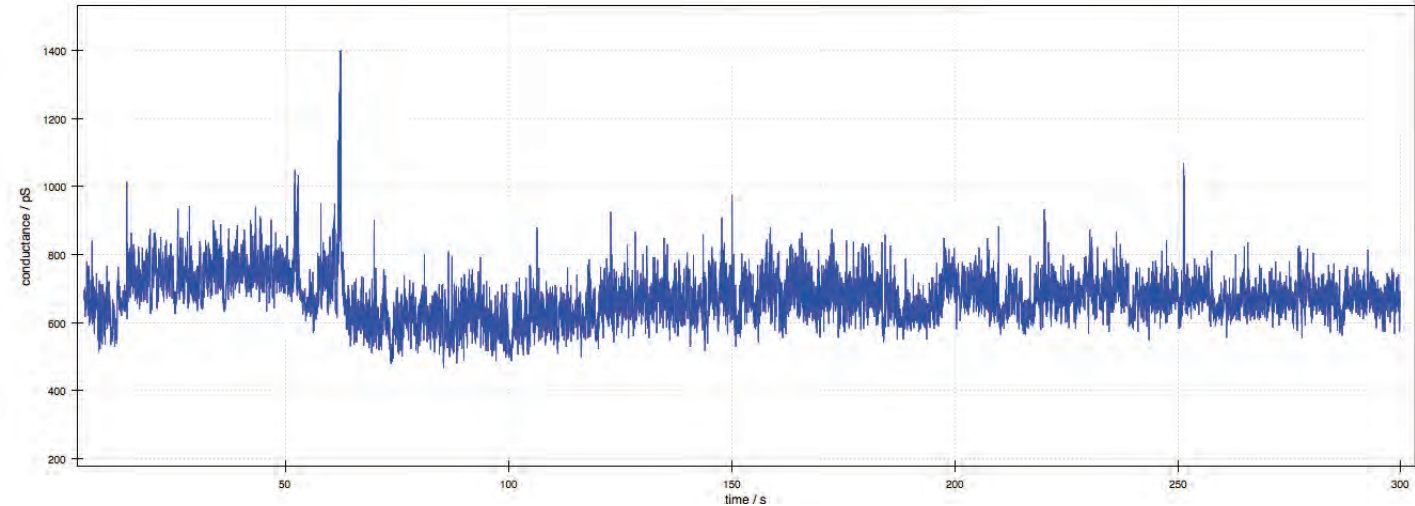


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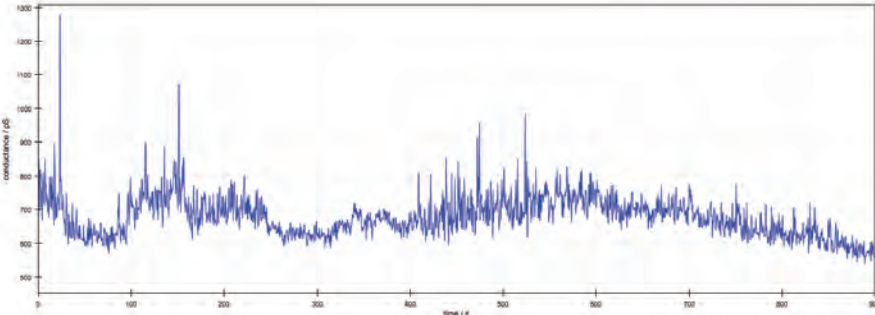
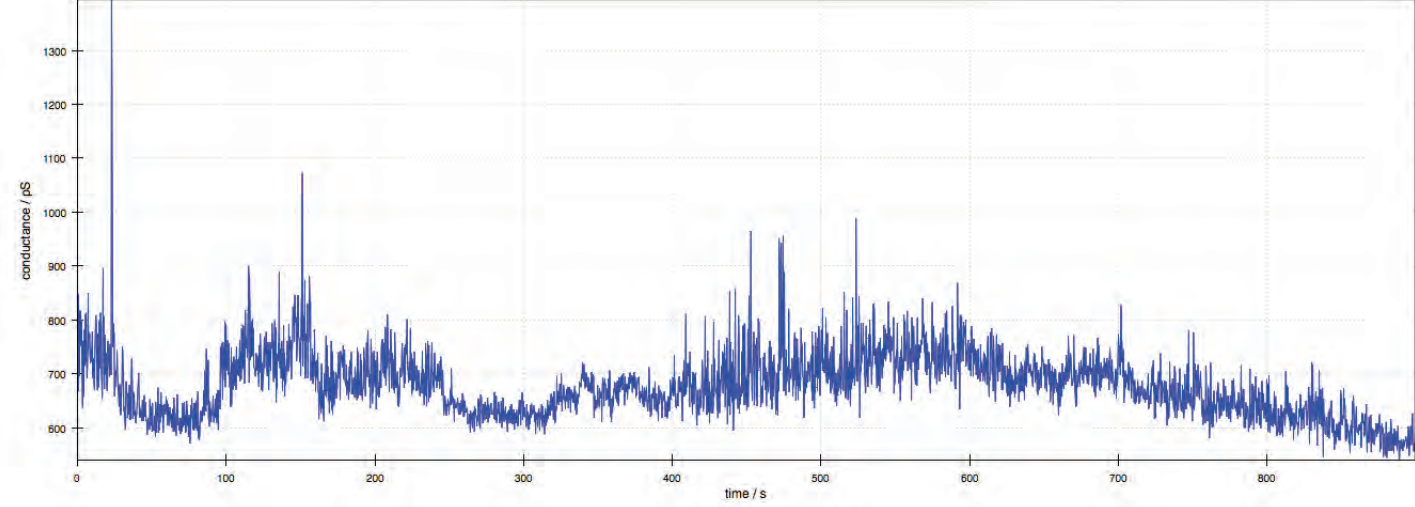




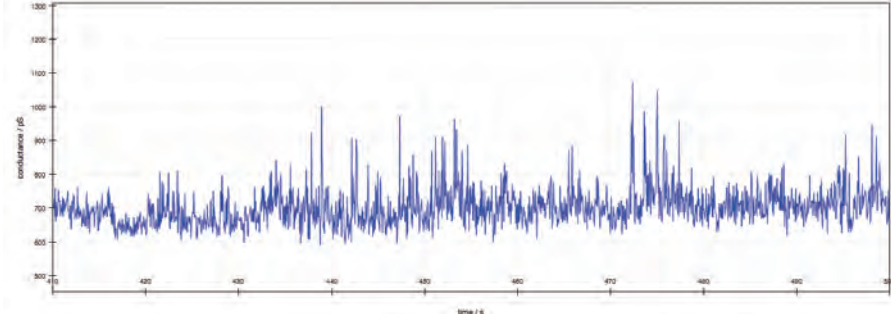
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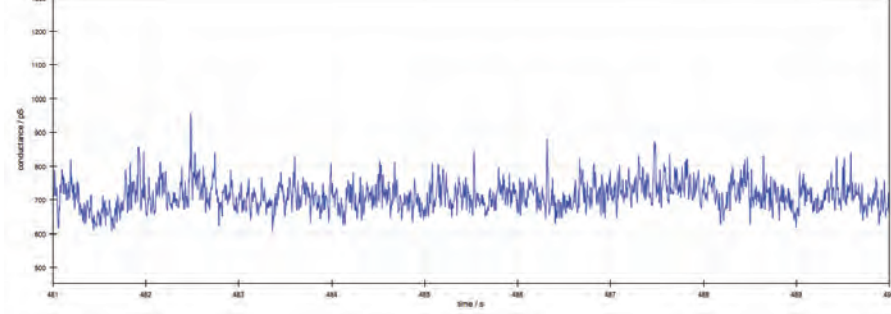
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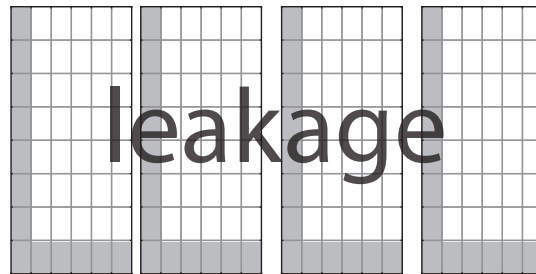
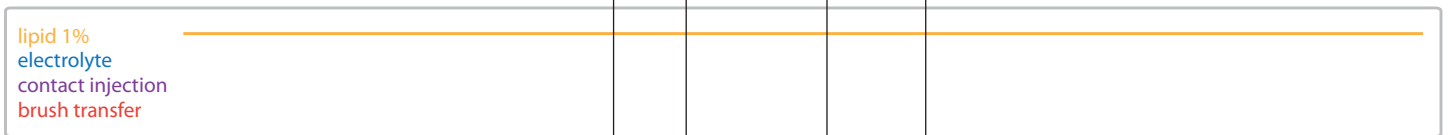
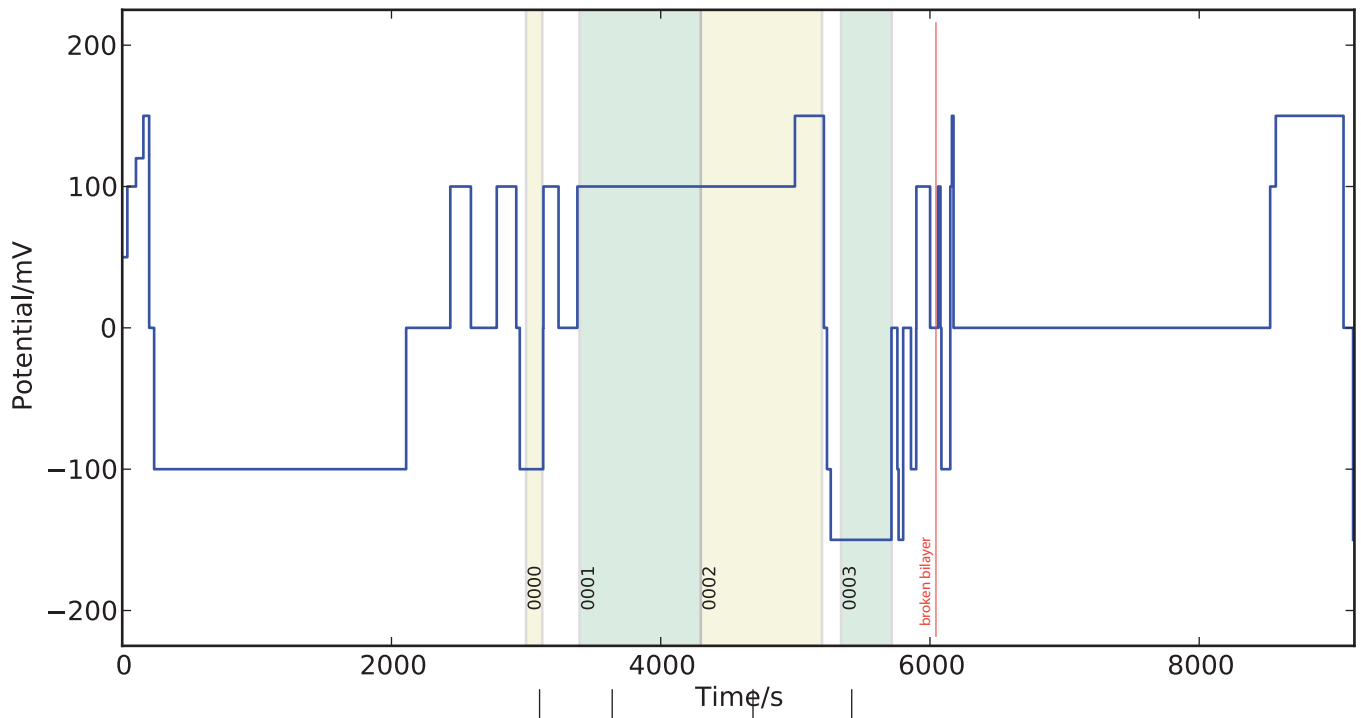


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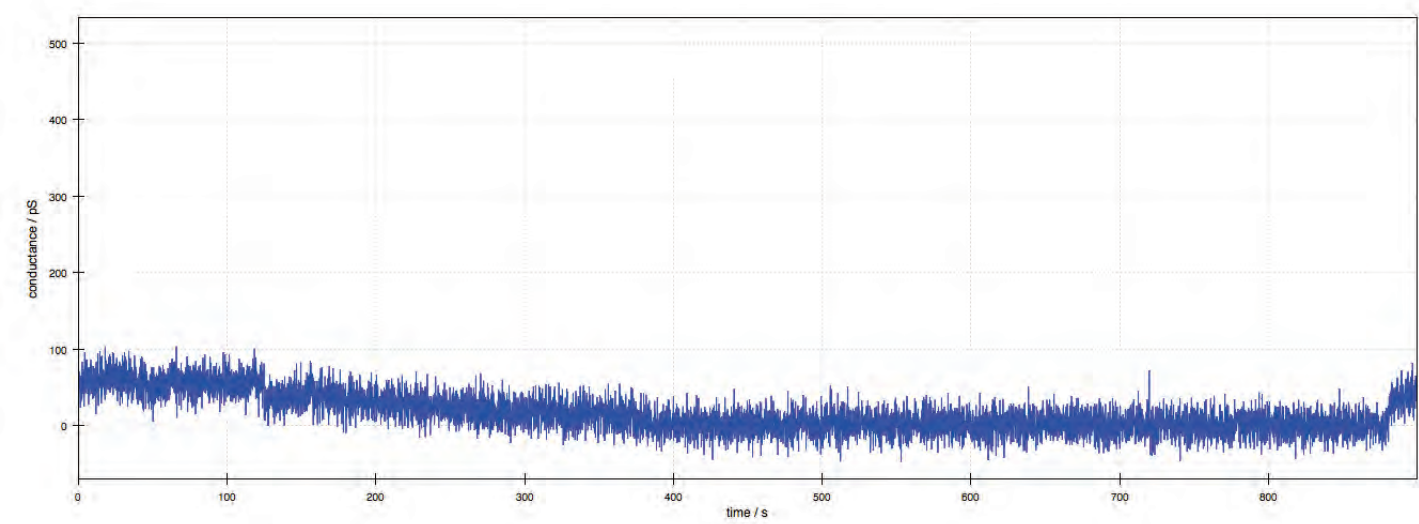


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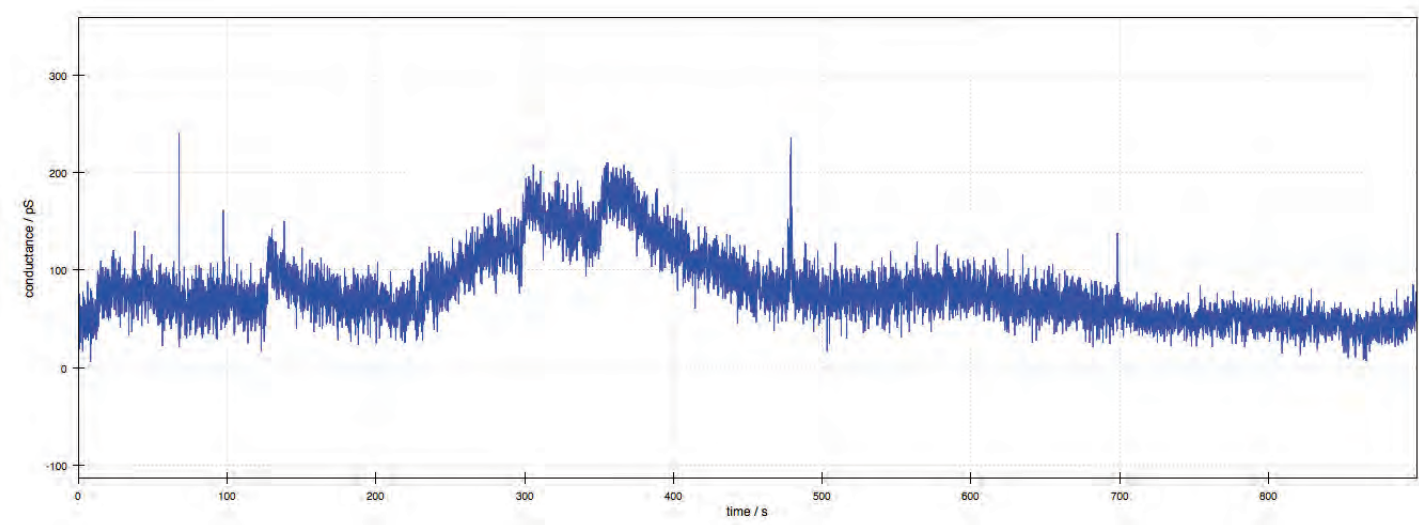
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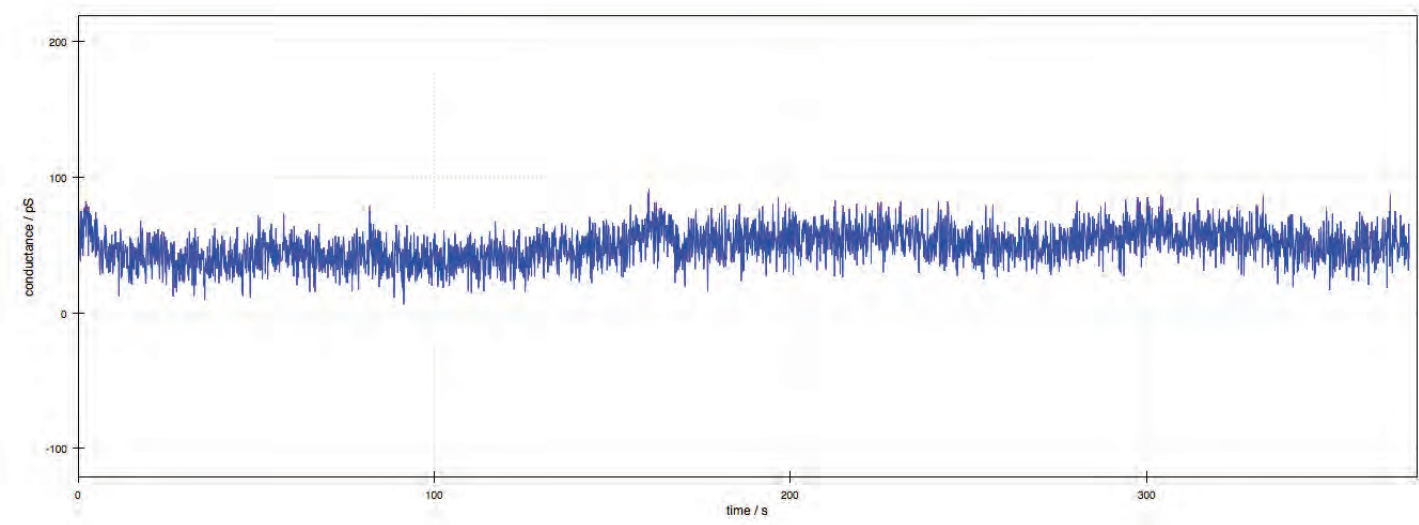
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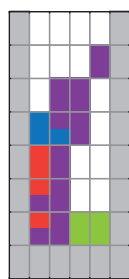
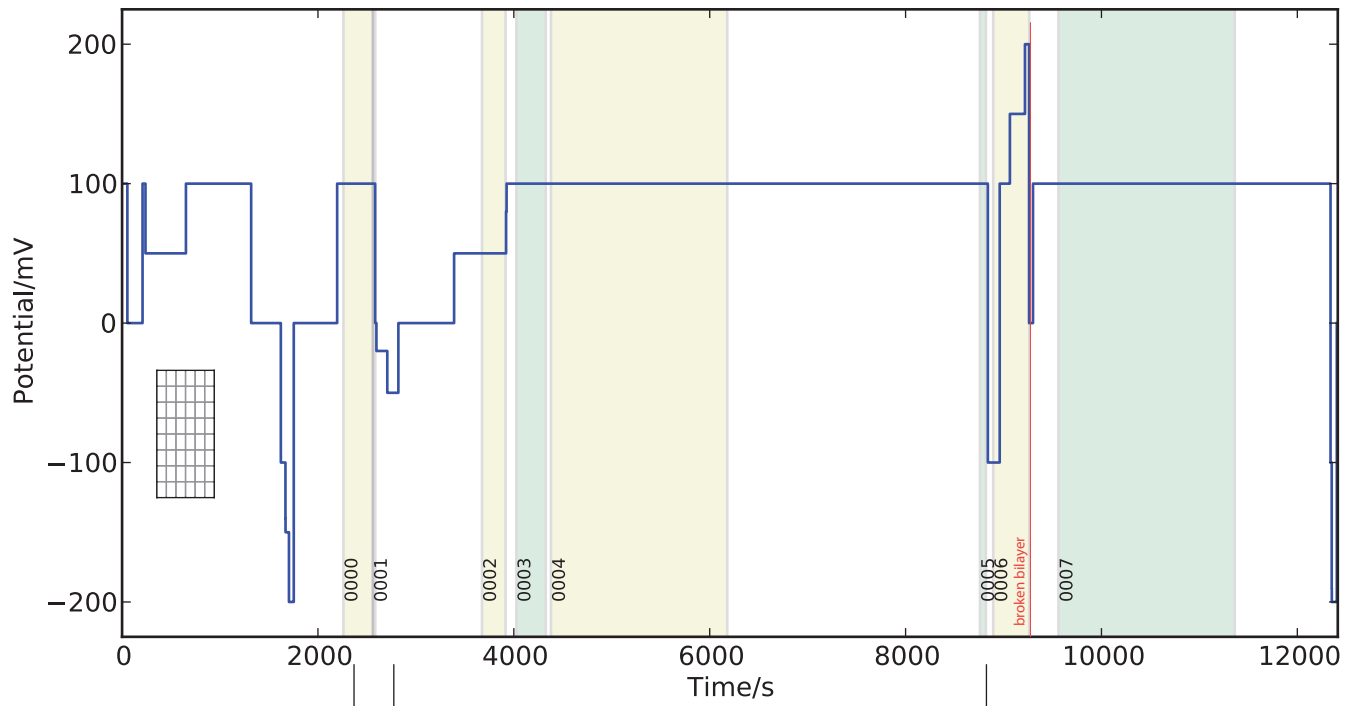
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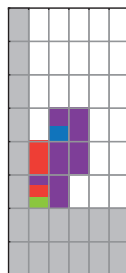
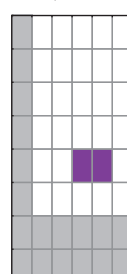
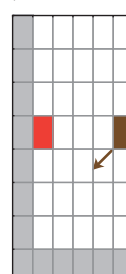
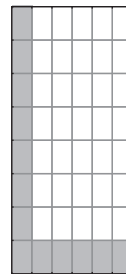
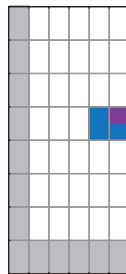
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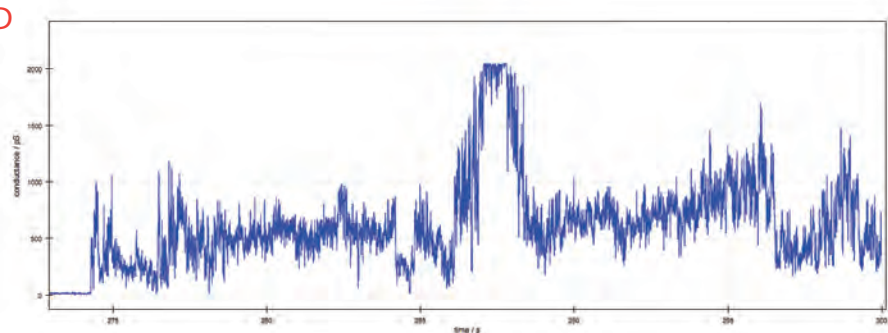
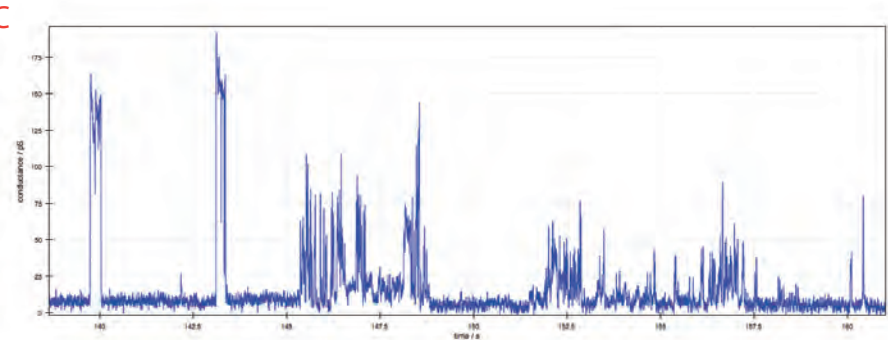
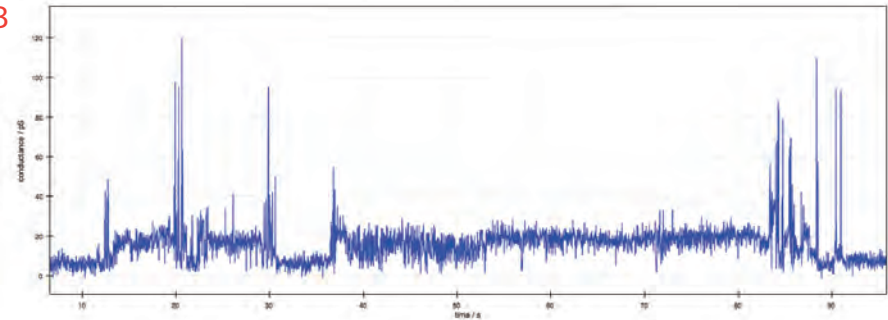
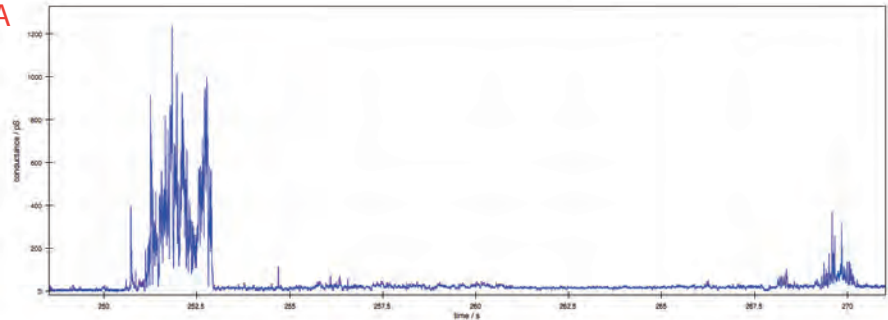
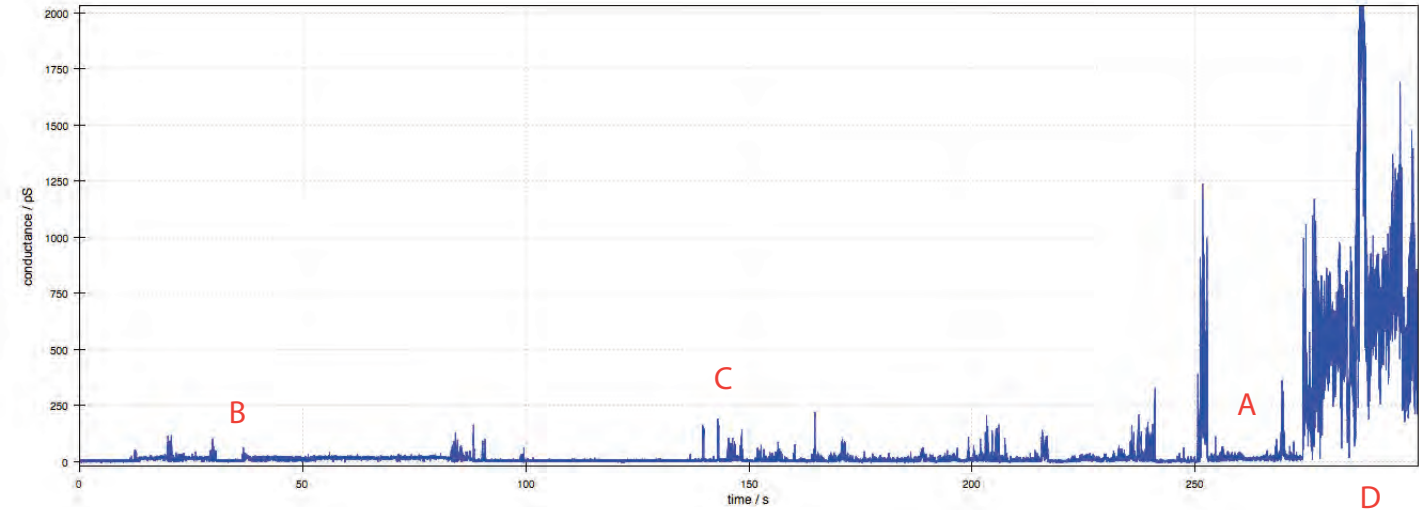
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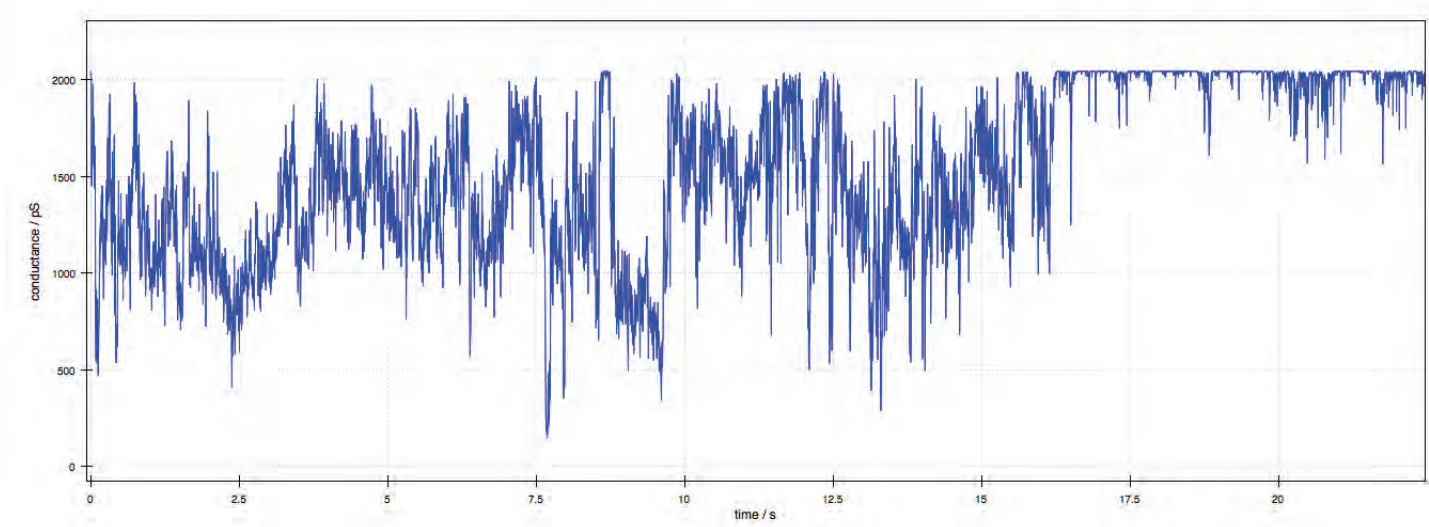
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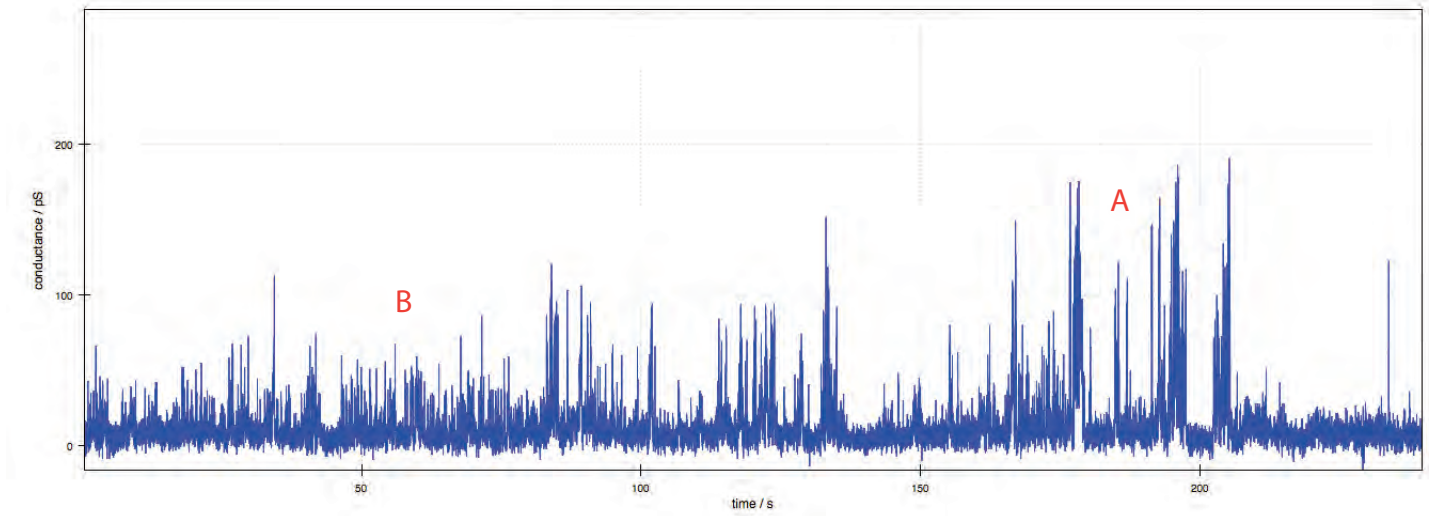
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but not entirely clean by NMR



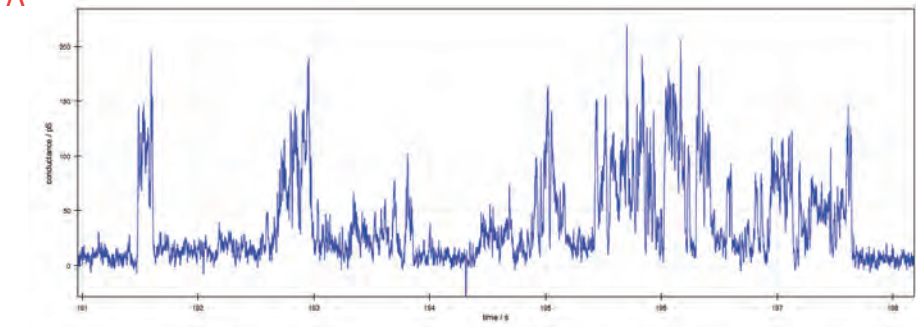
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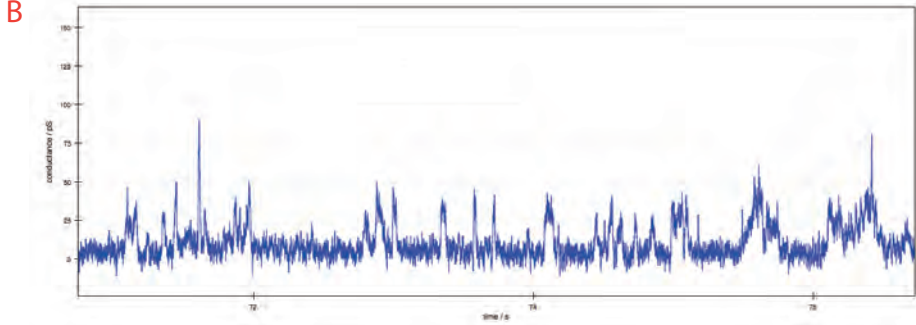
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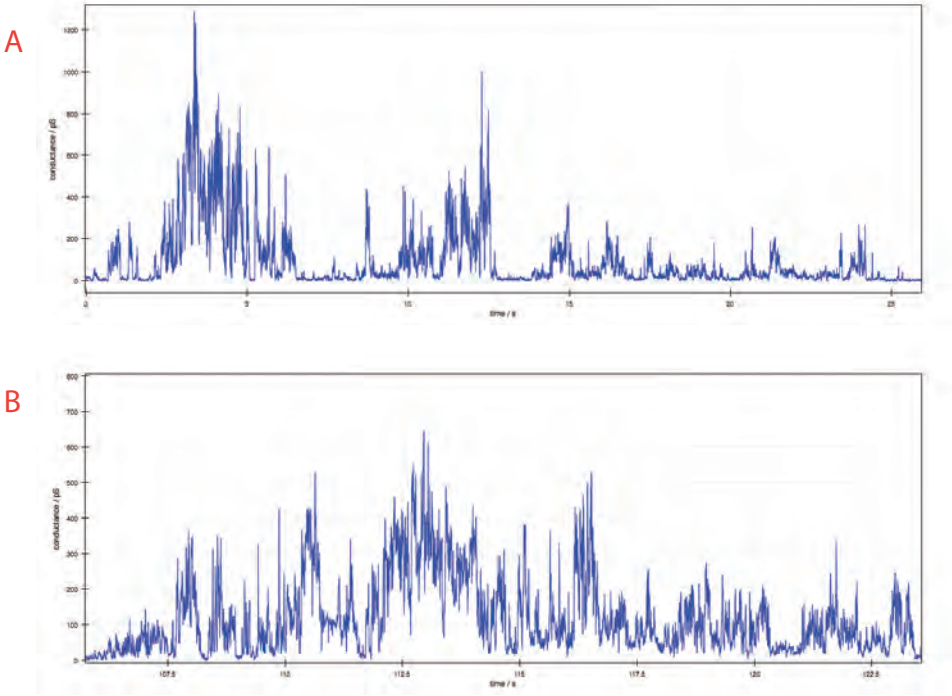
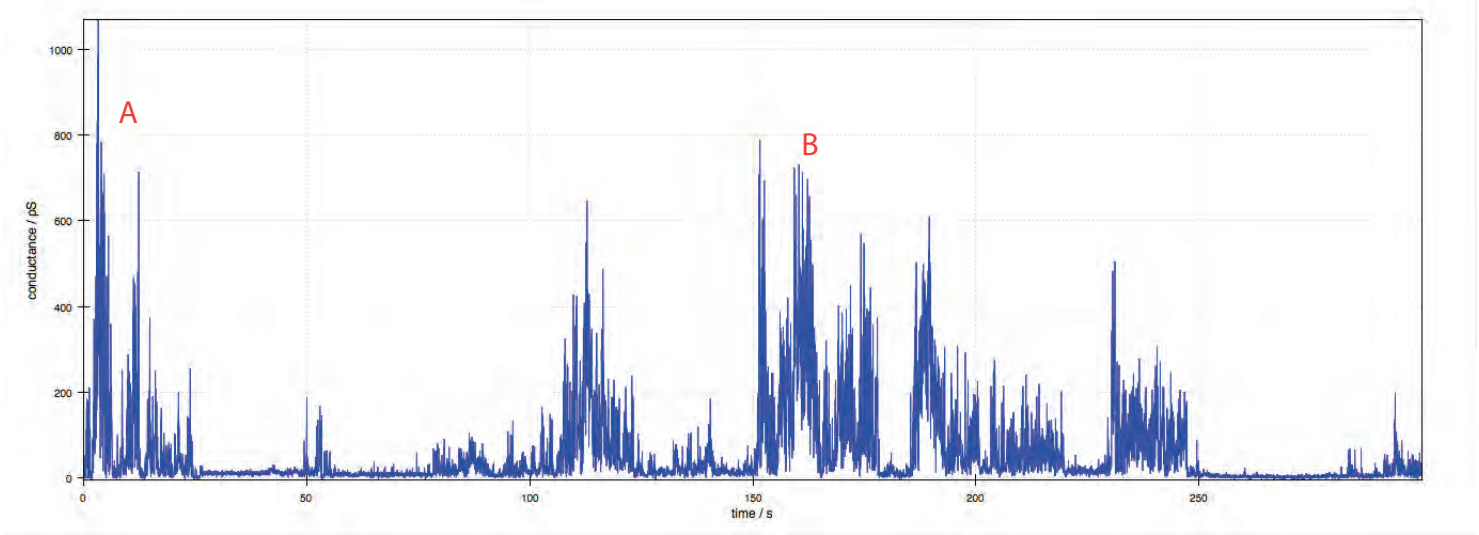
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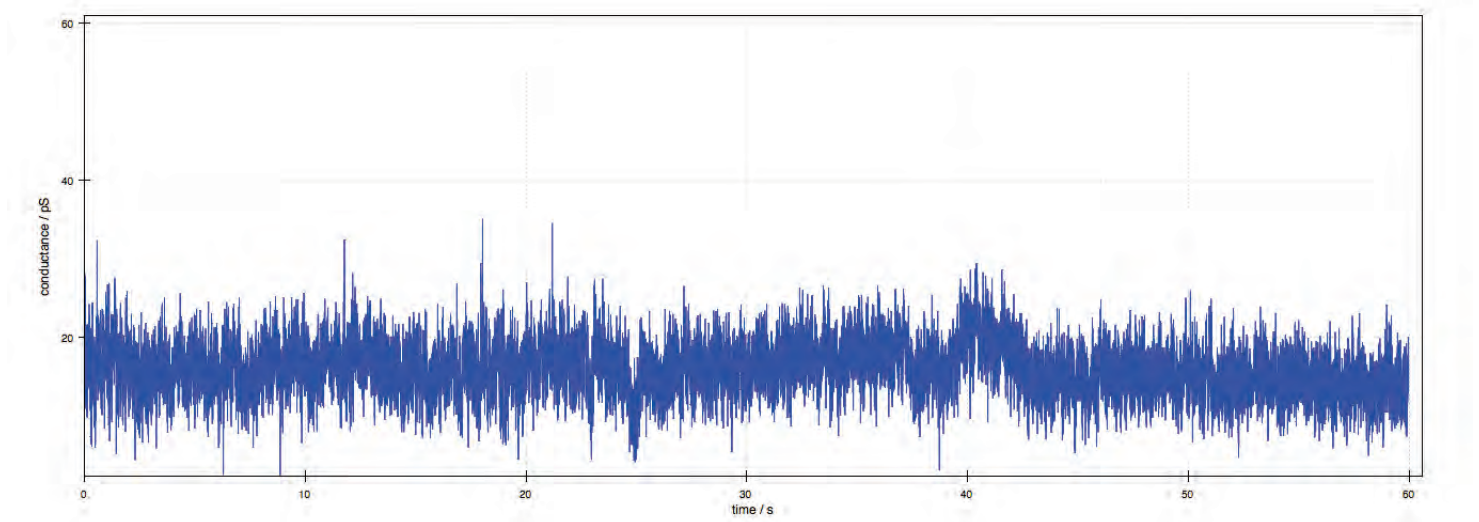
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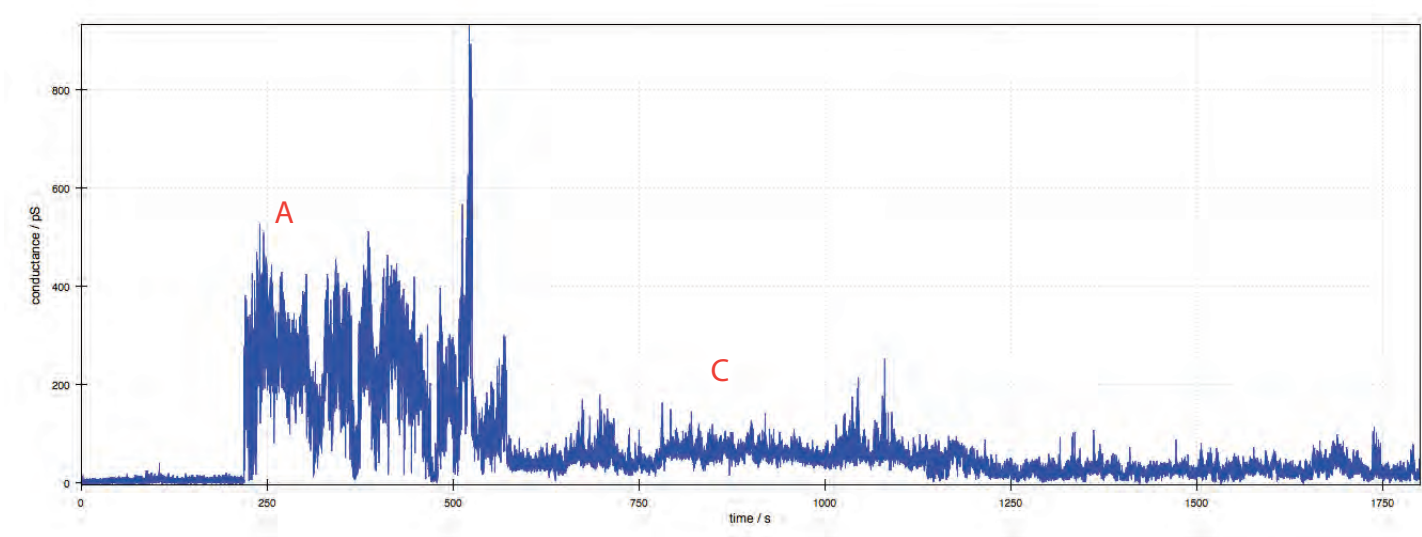


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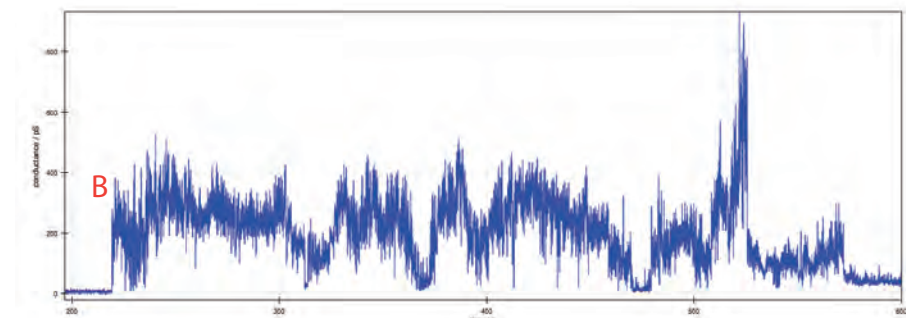


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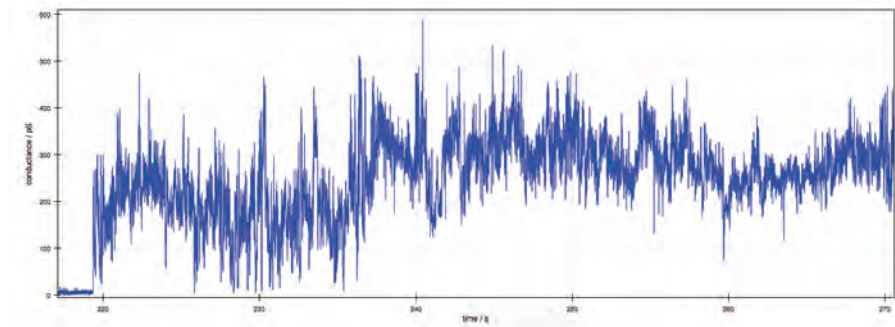




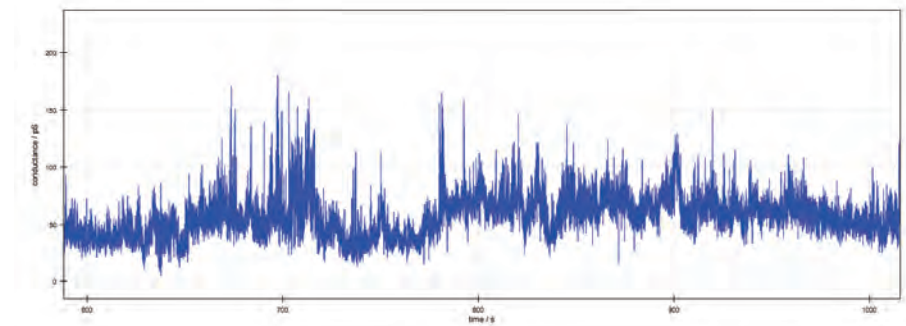
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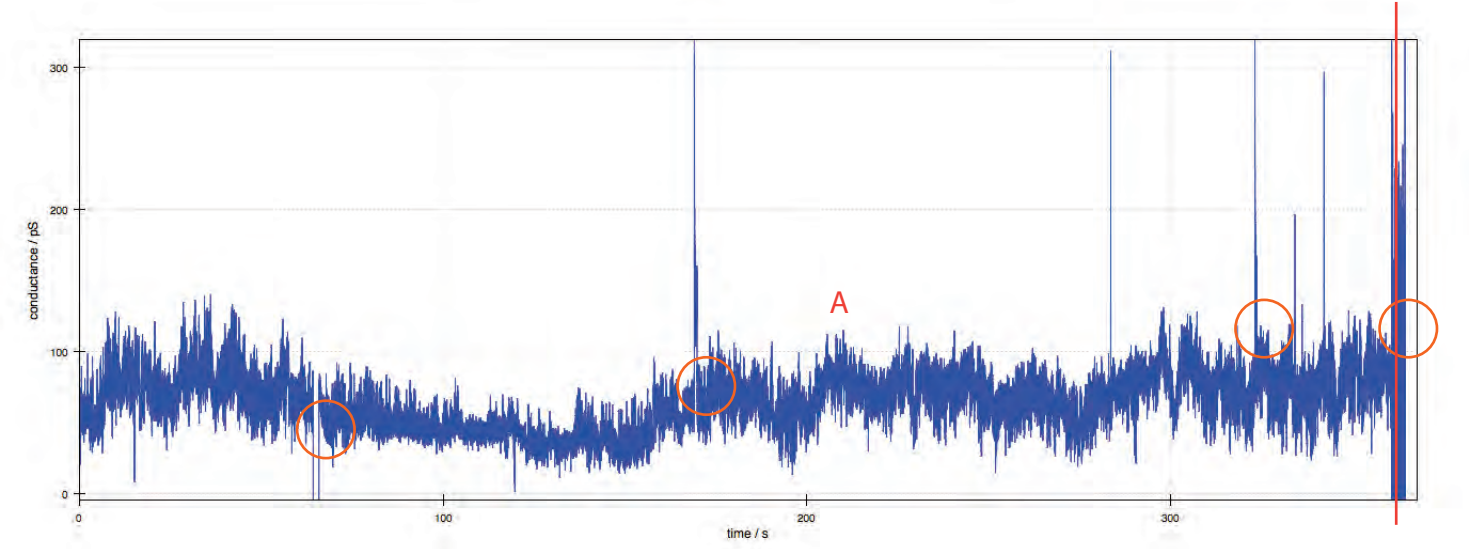
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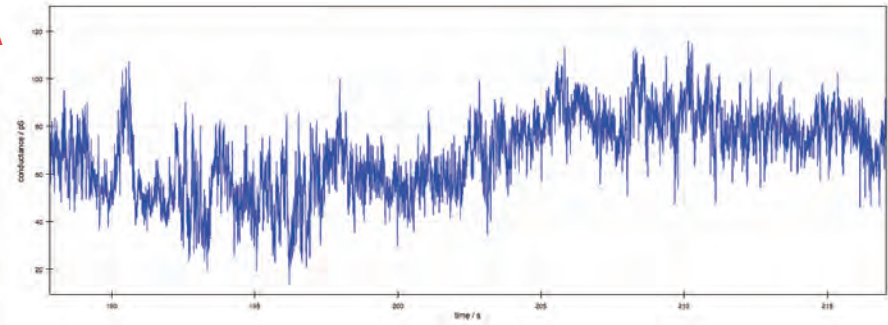
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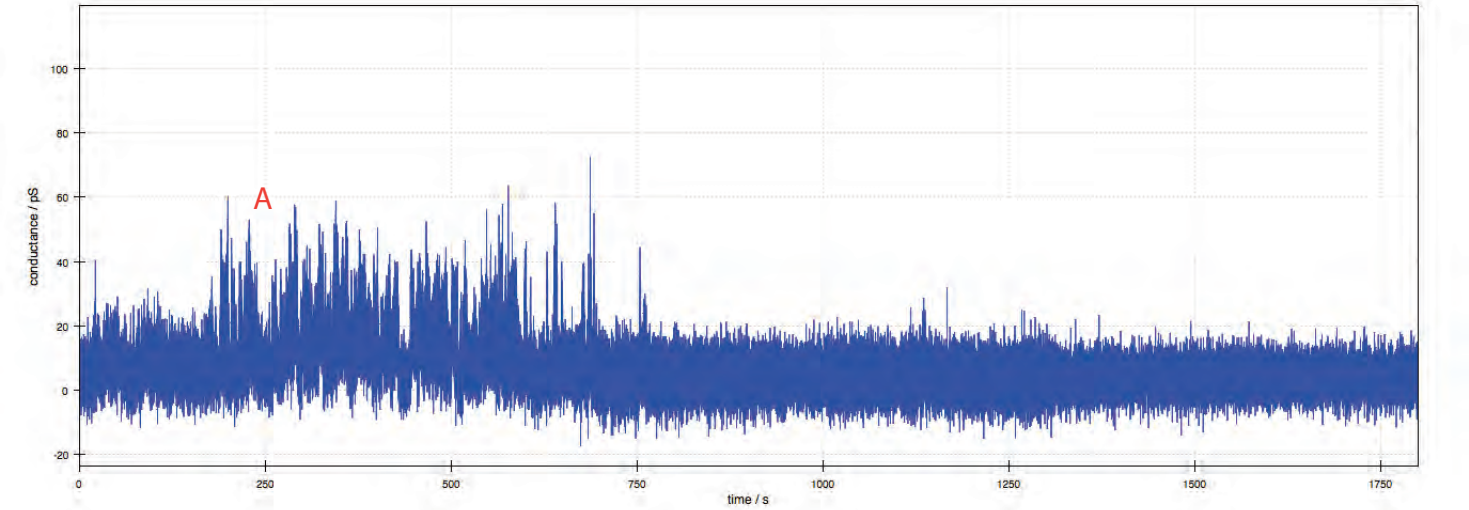
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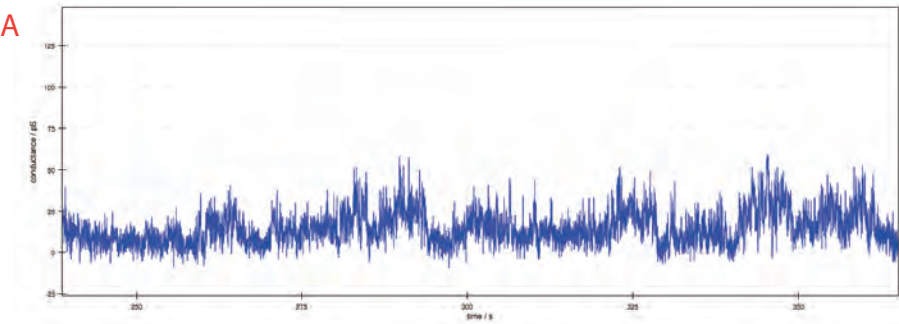
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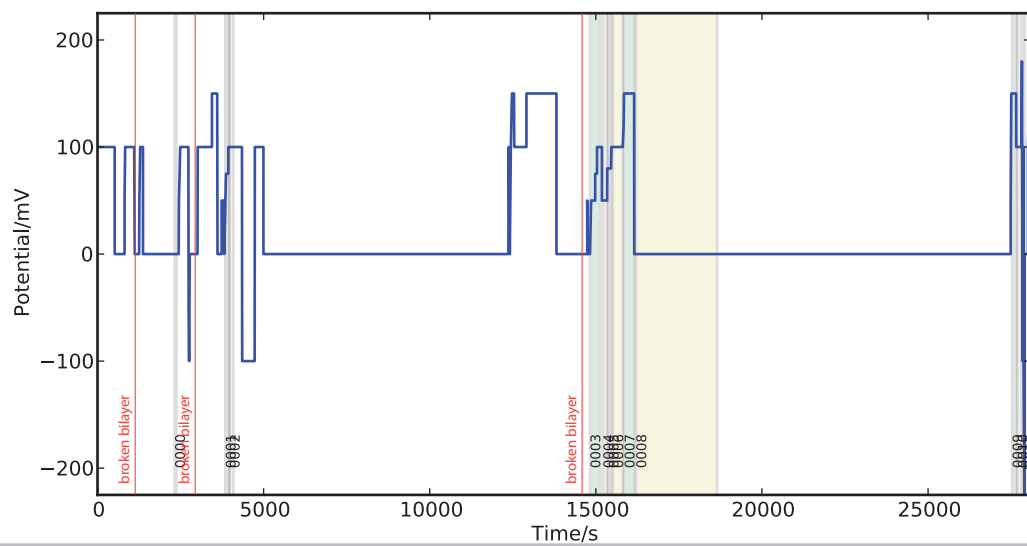
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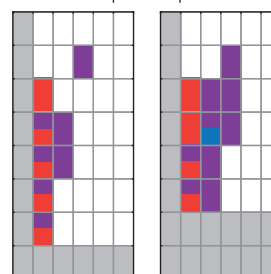
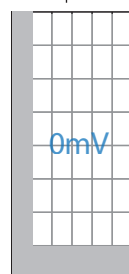
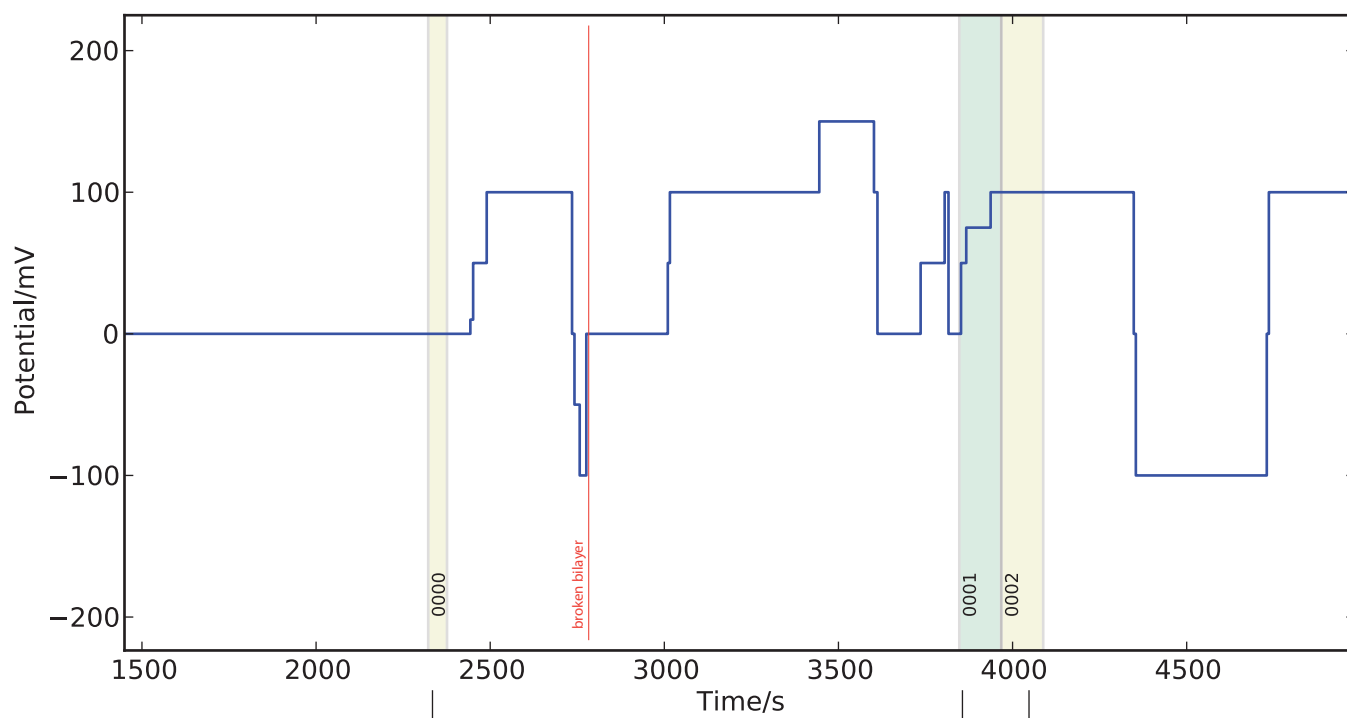
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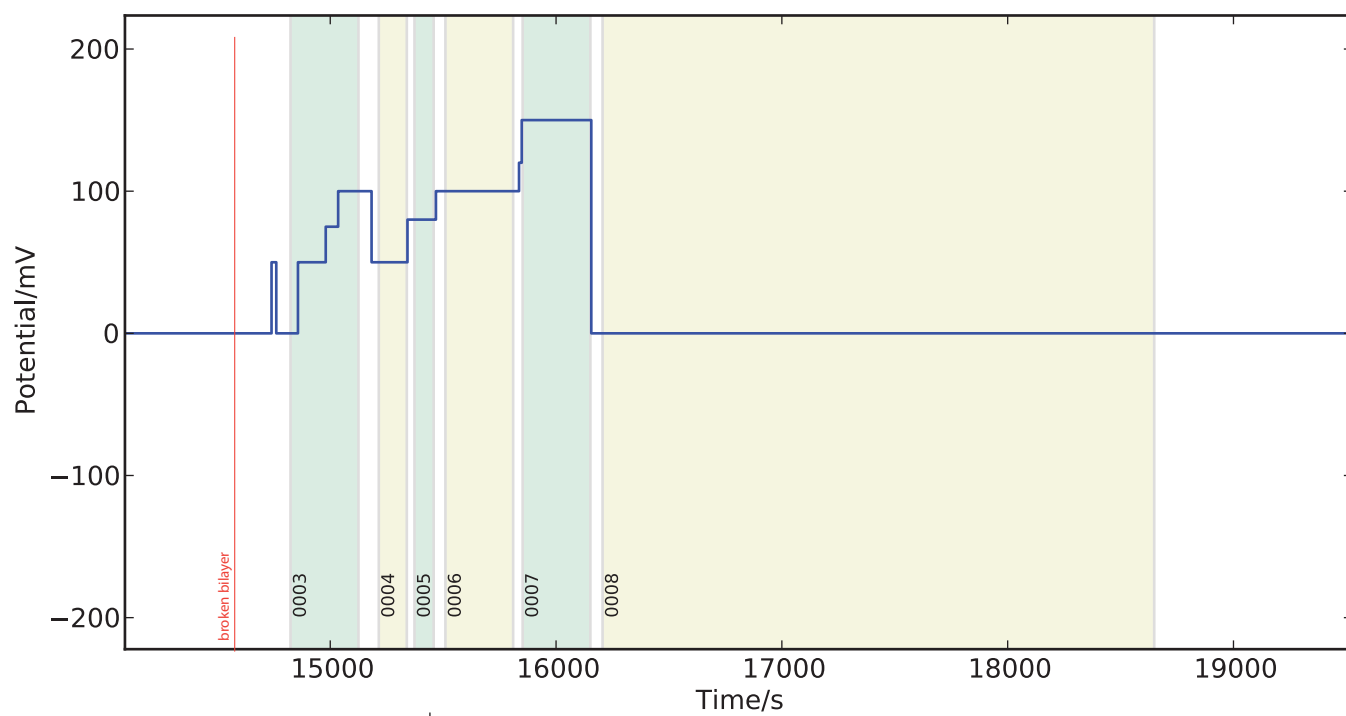
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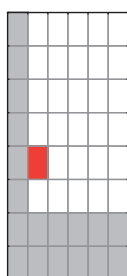
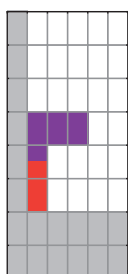
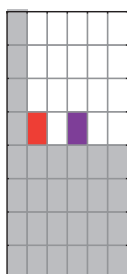
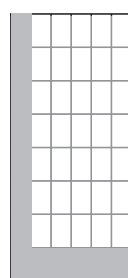
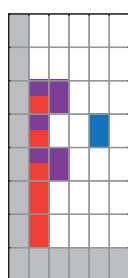
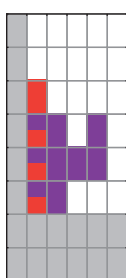
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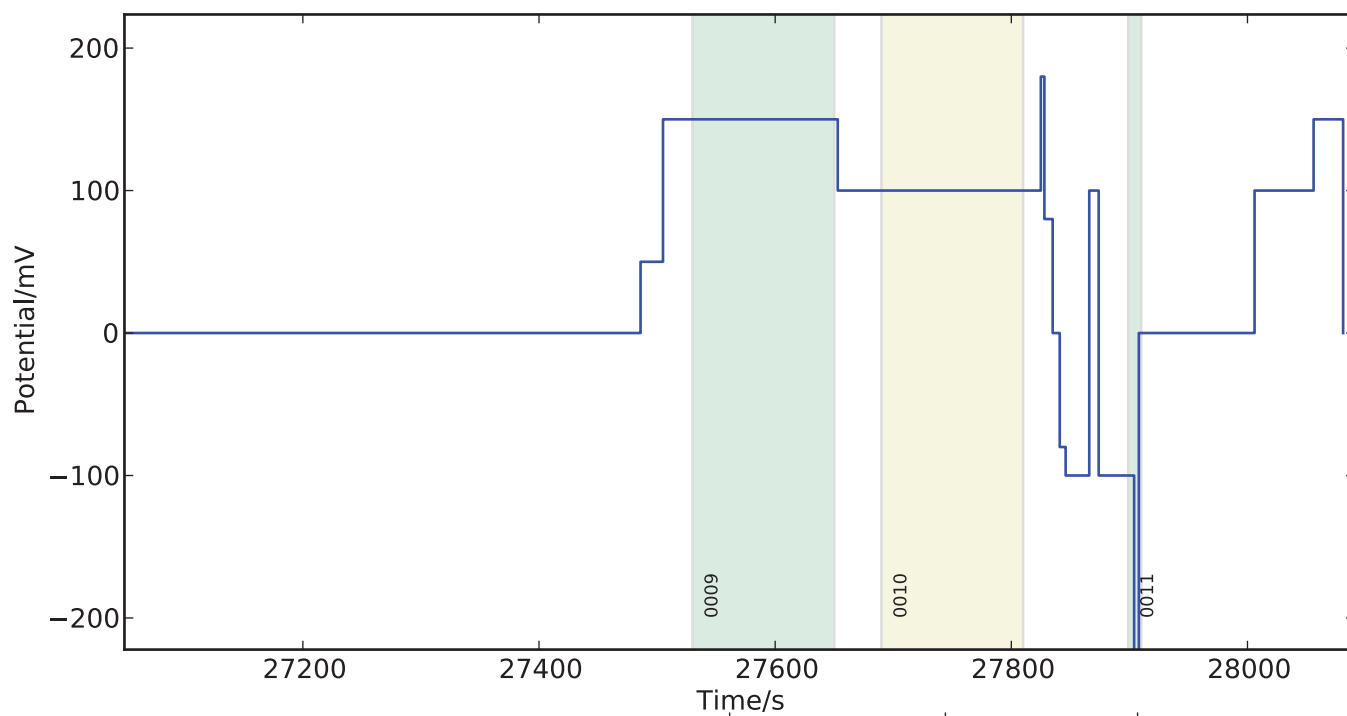
JC-124 bCDMAmAm3O1 - JC-V-1 1M CsCl pH 7



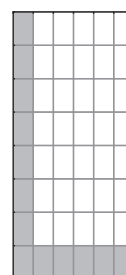
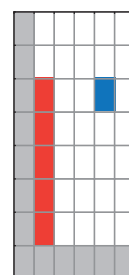
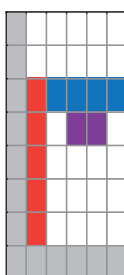
lipid
electrolyte
contact injection 33/20uM
brush transfer



JC-124 bCDMaIAm3O1 - JC-V-1 1M CsCl pH 7

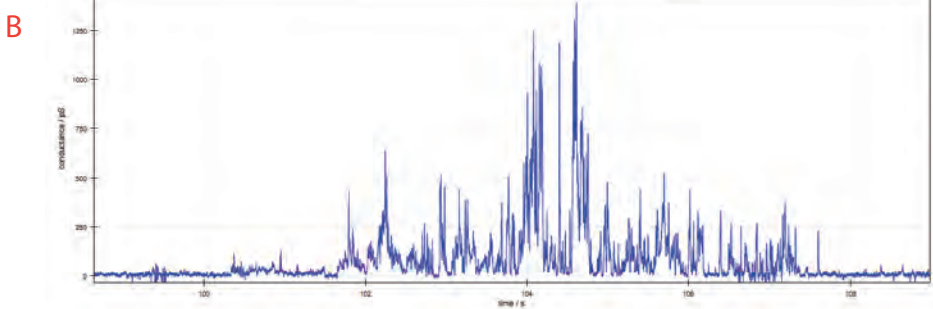
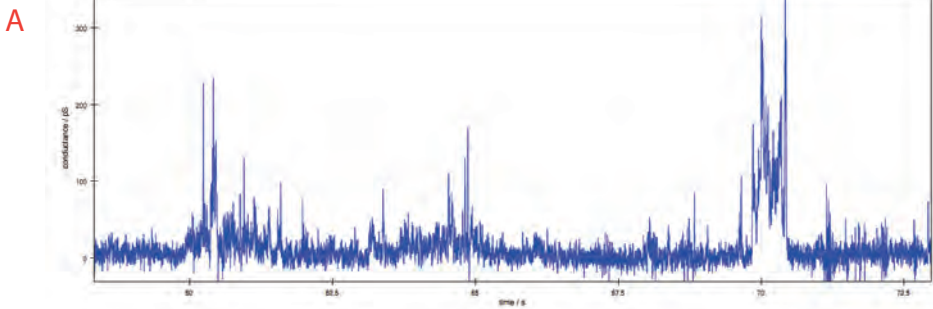
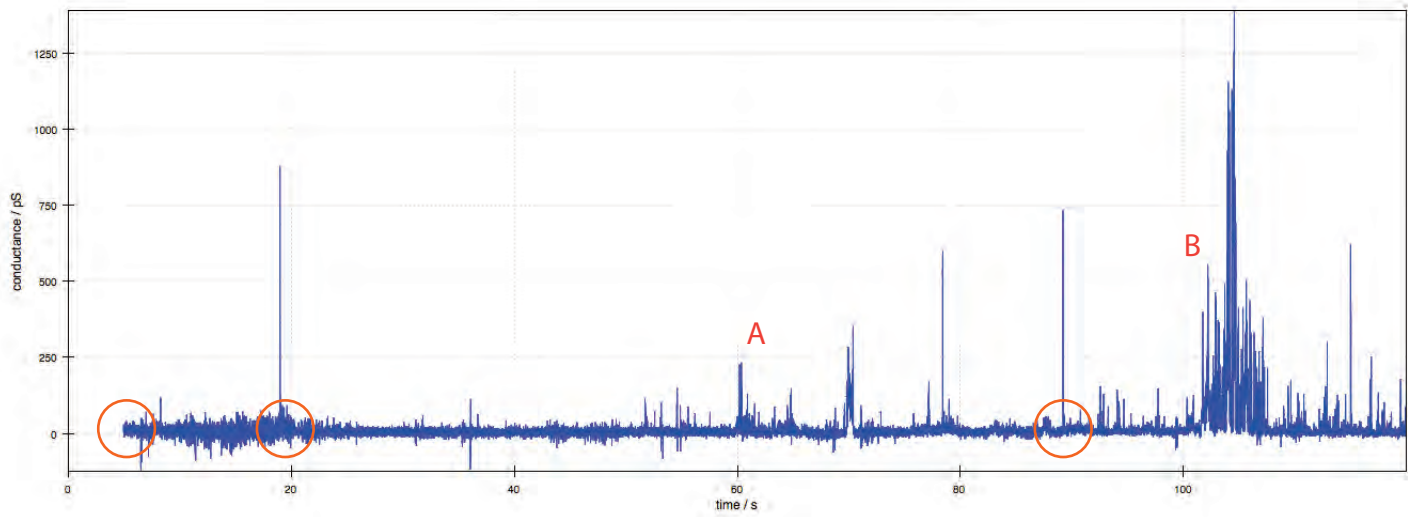


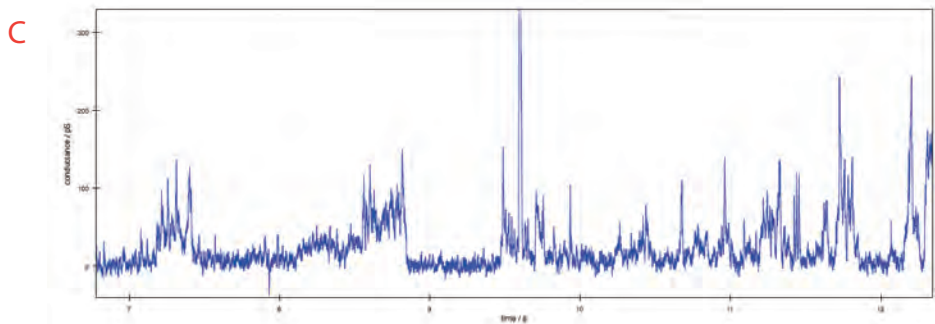
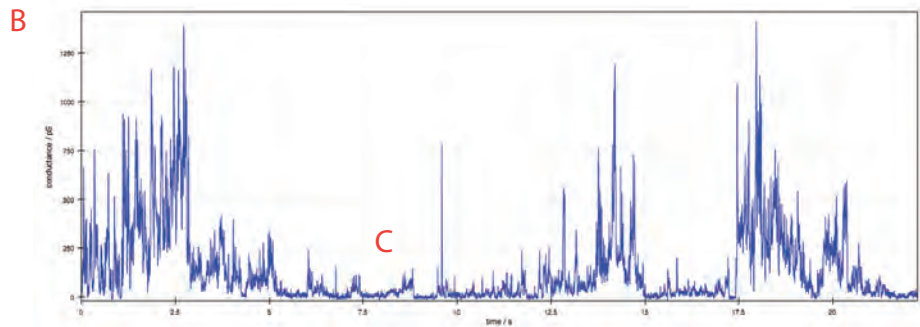
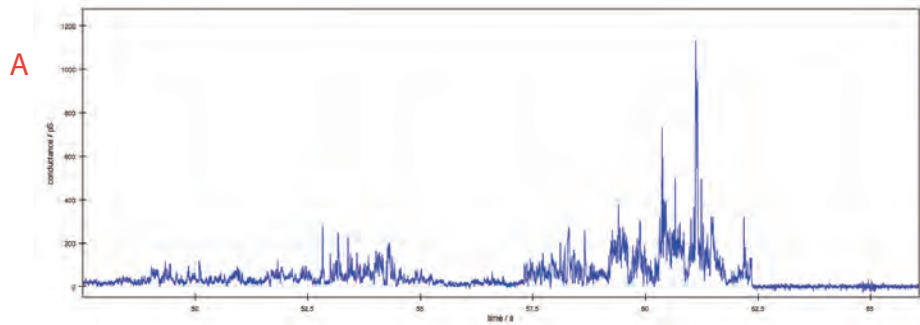
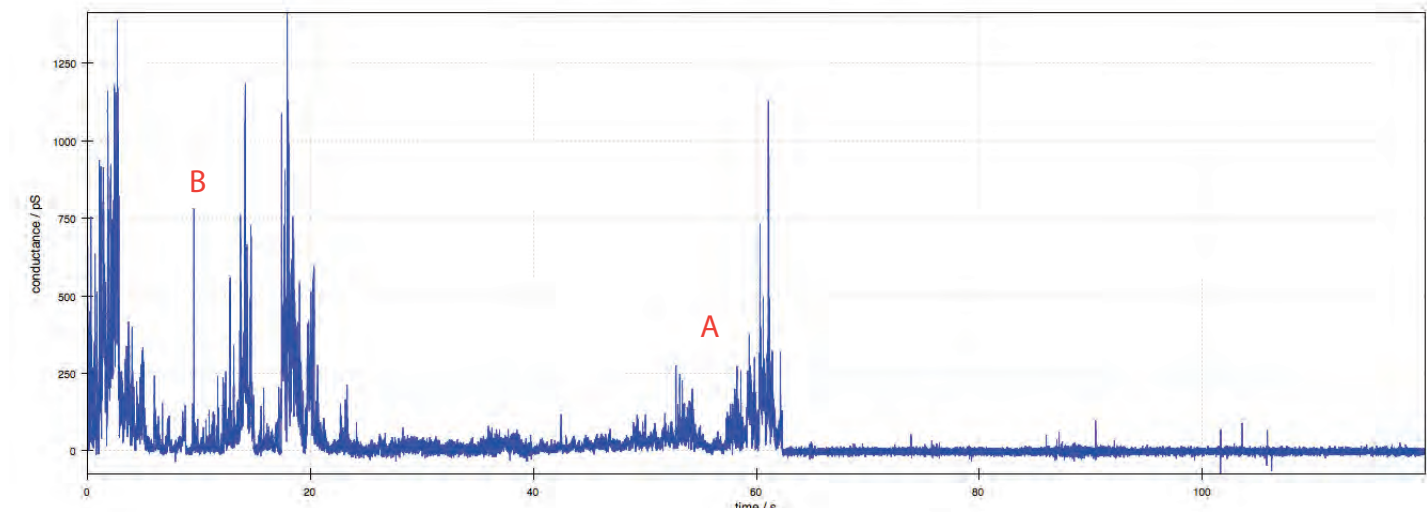
lipid
electrolyte
contact injection 33/20uM
brush transfer



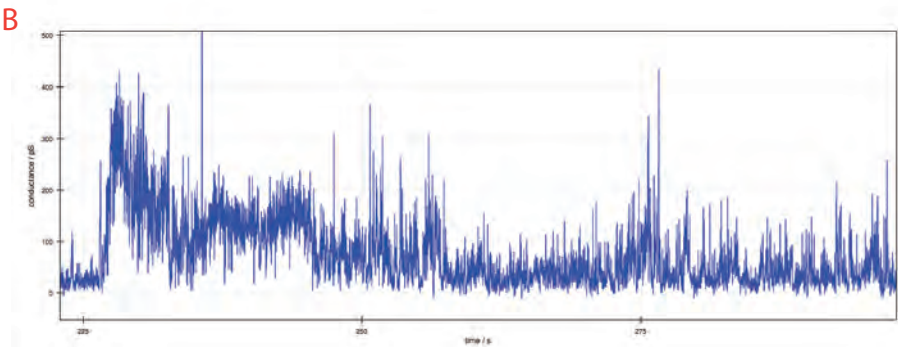
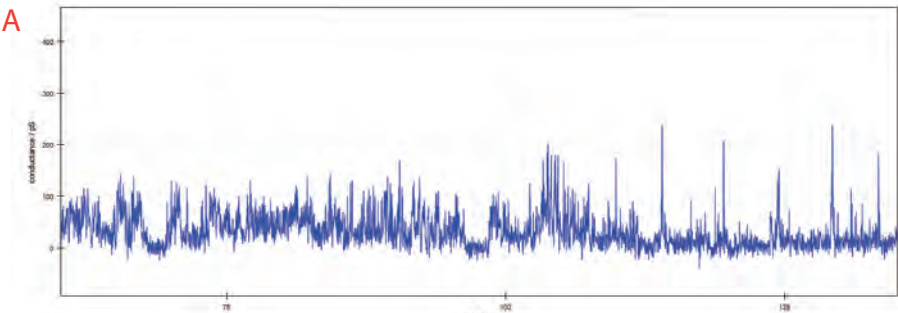
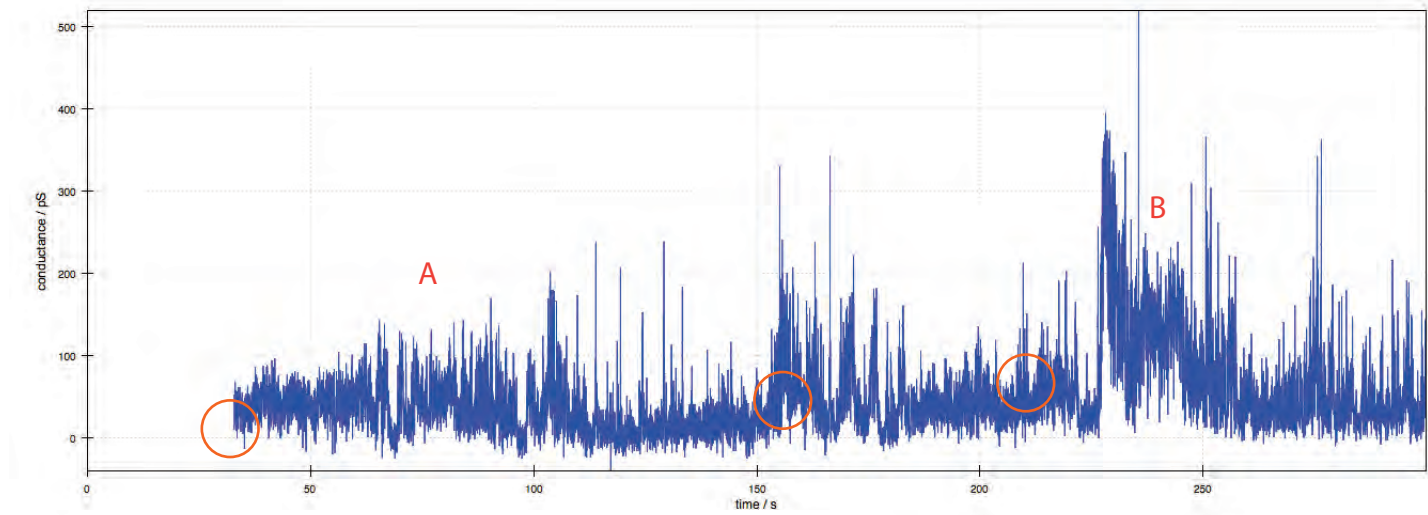
asymmetric?

124-0001

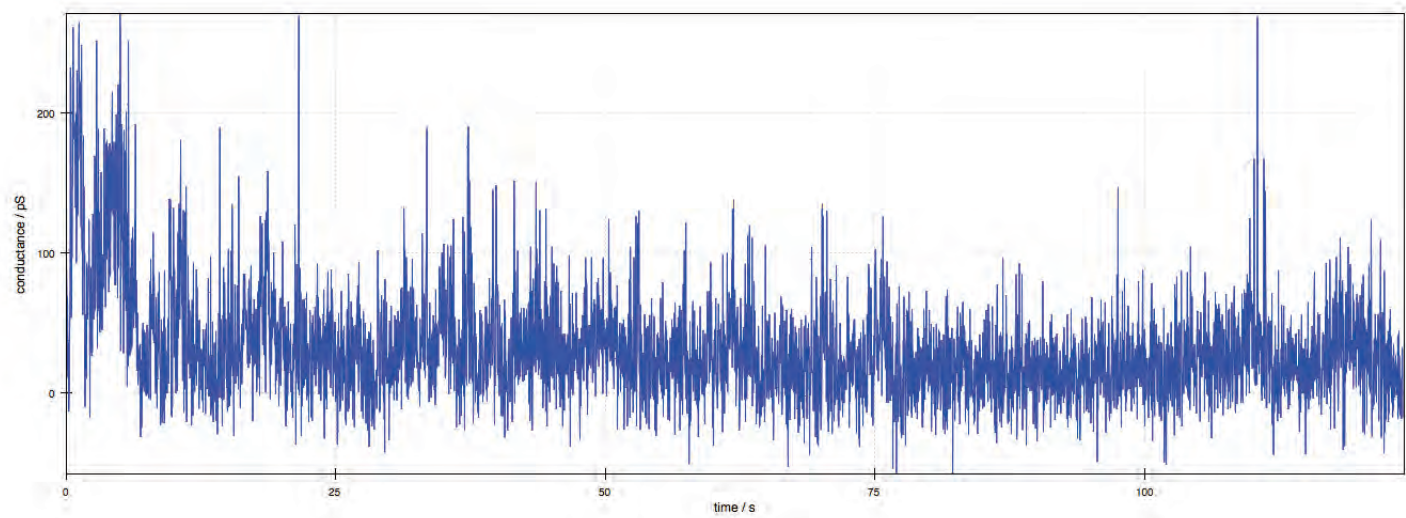




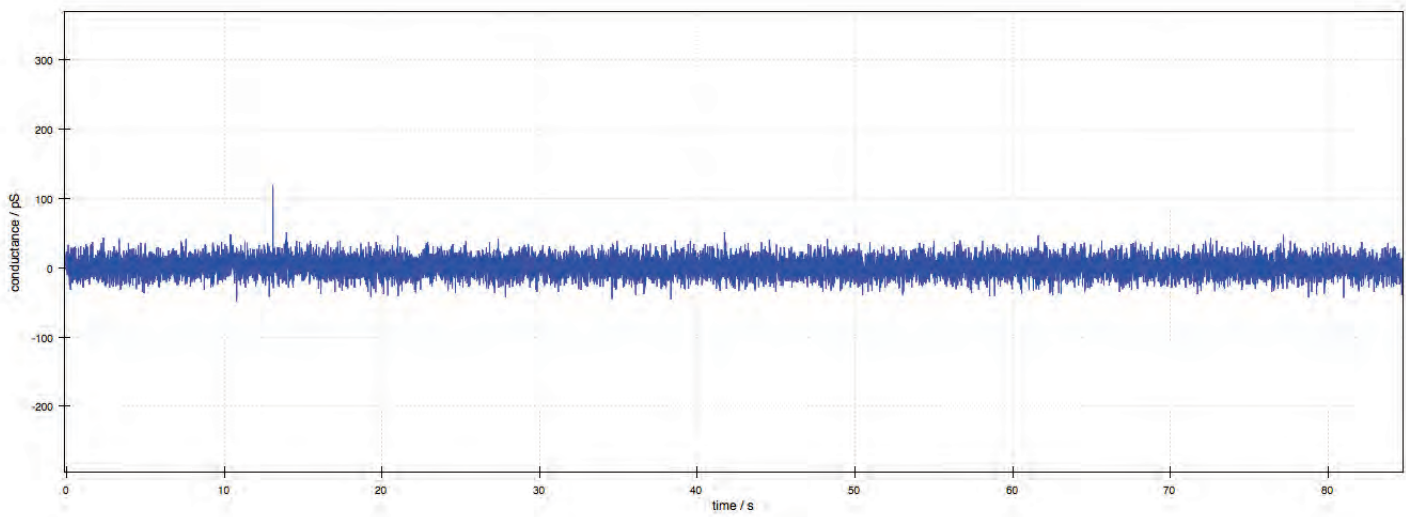
124-0003



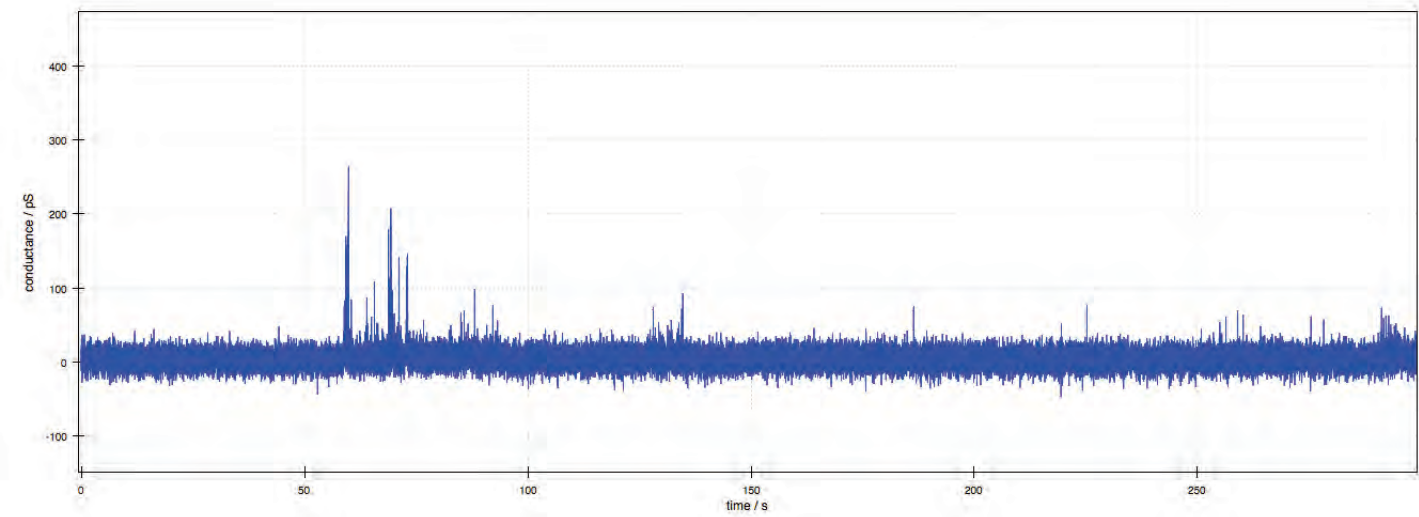
124-0004

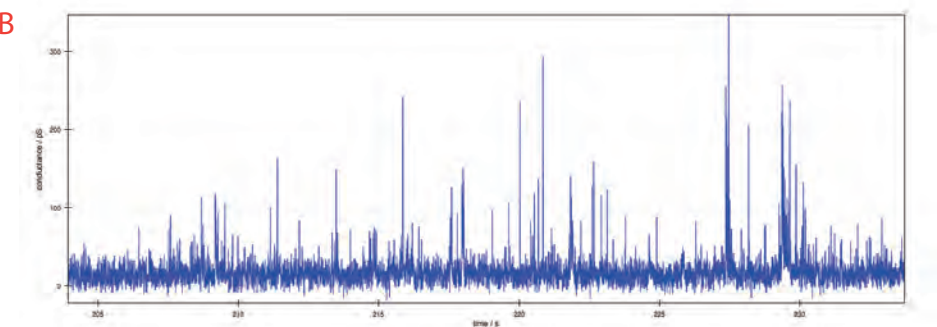
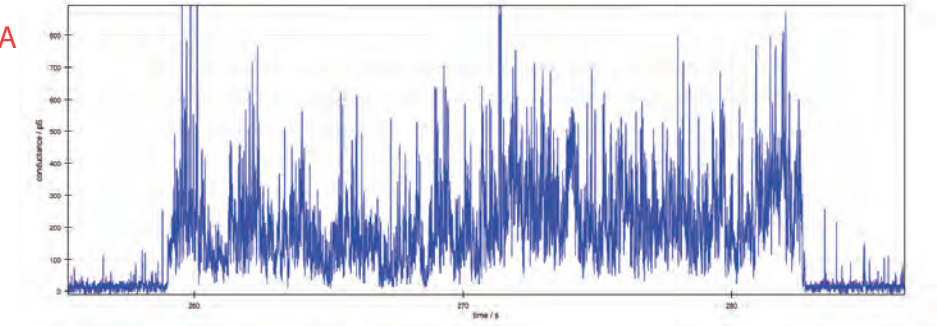
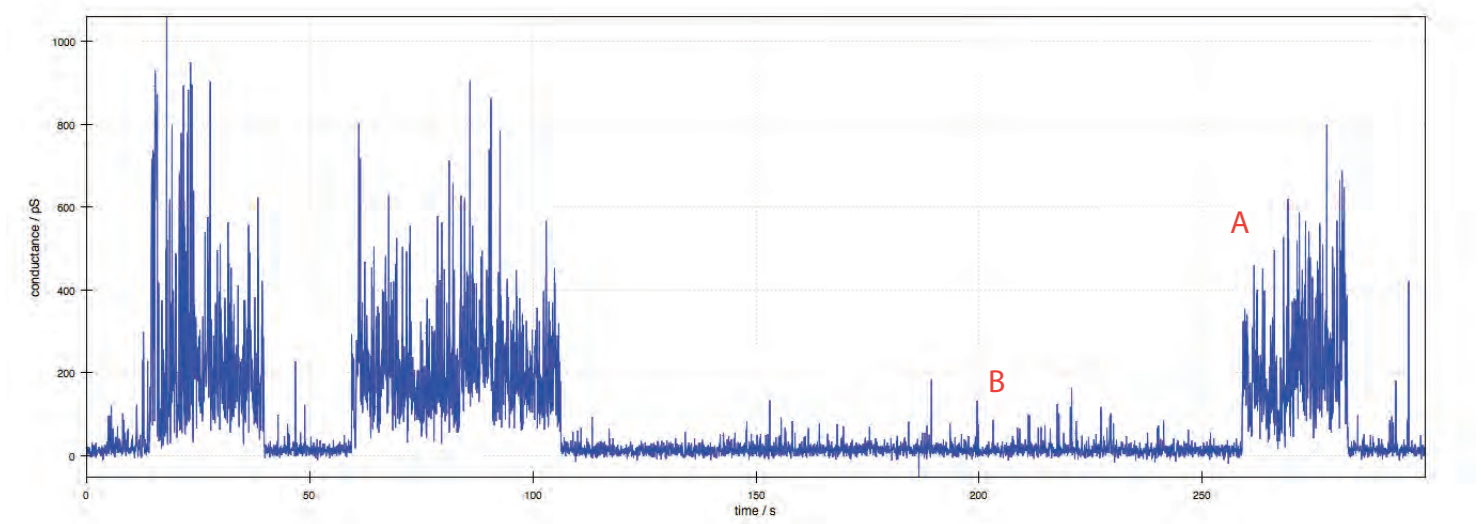


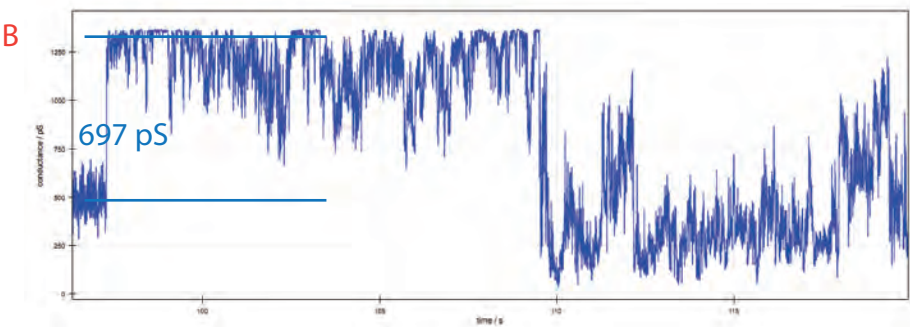
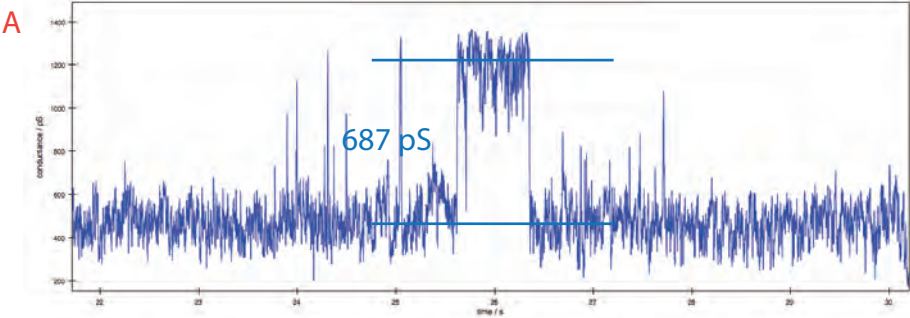
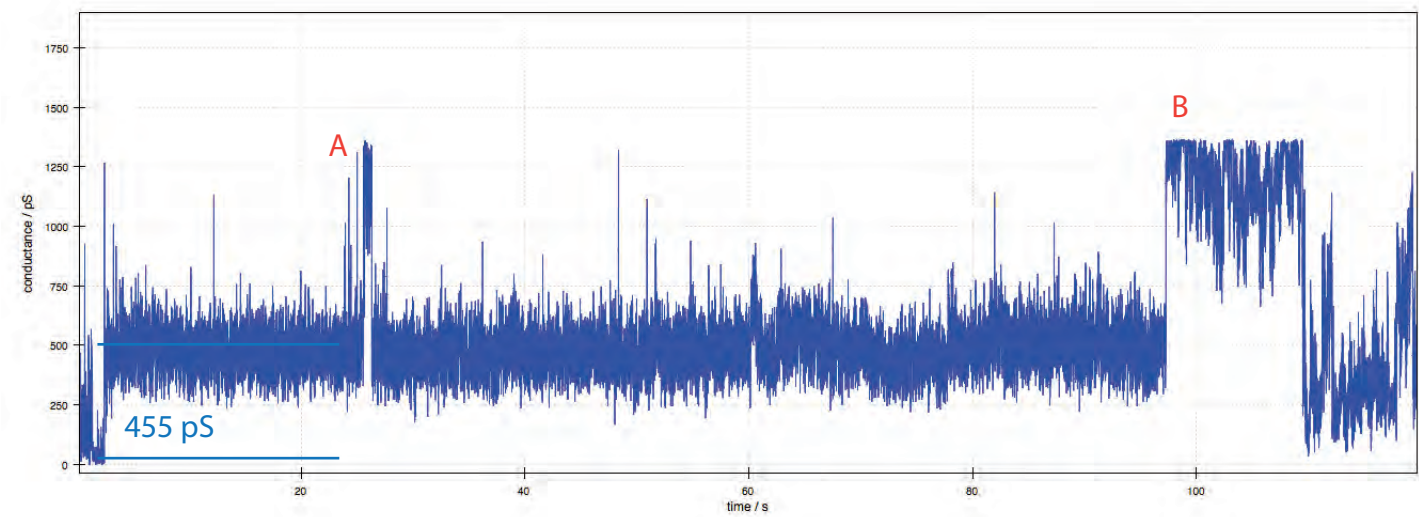
124-0005



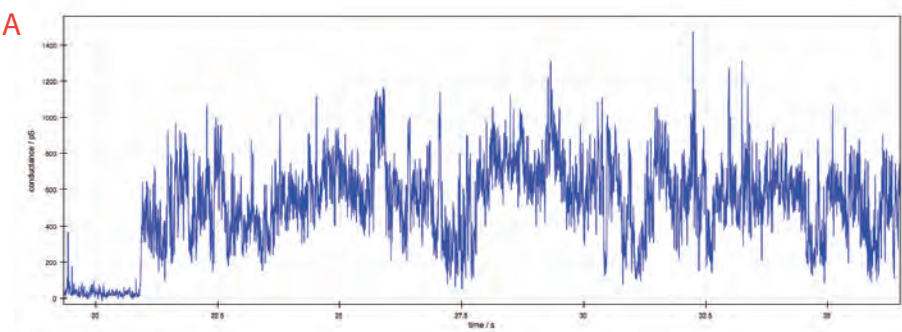
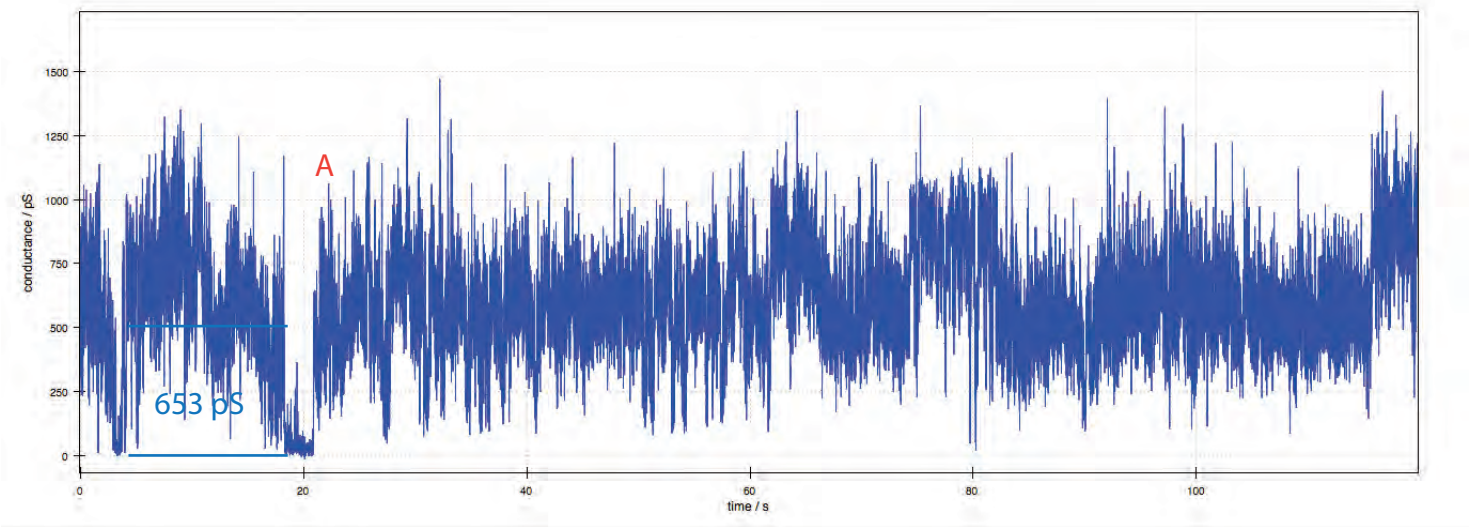
124-0006



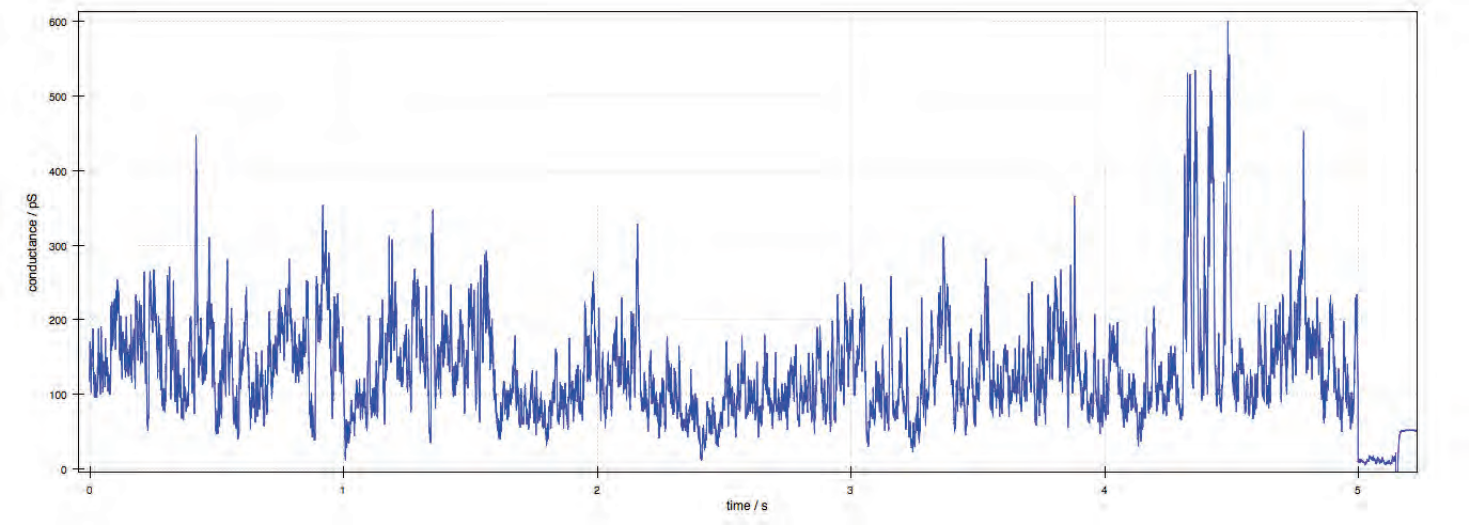




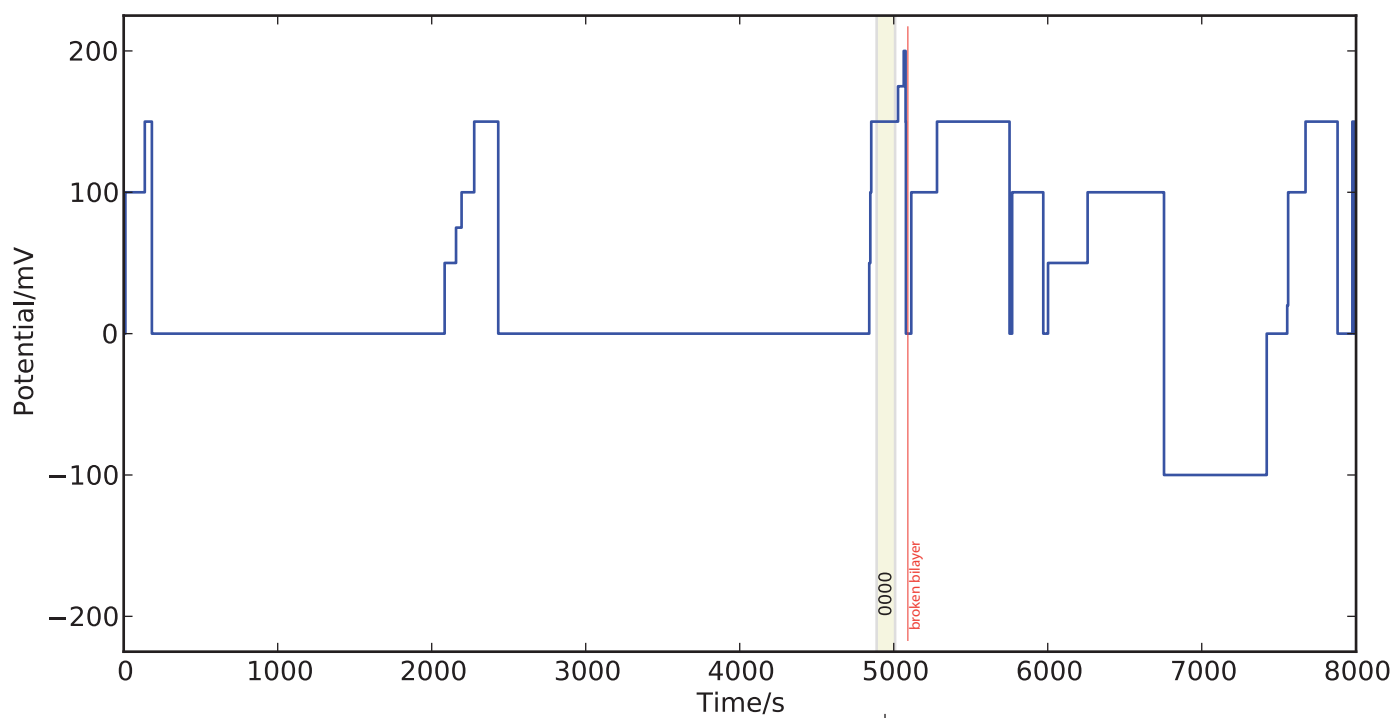
124-0010



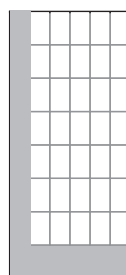
124-0011

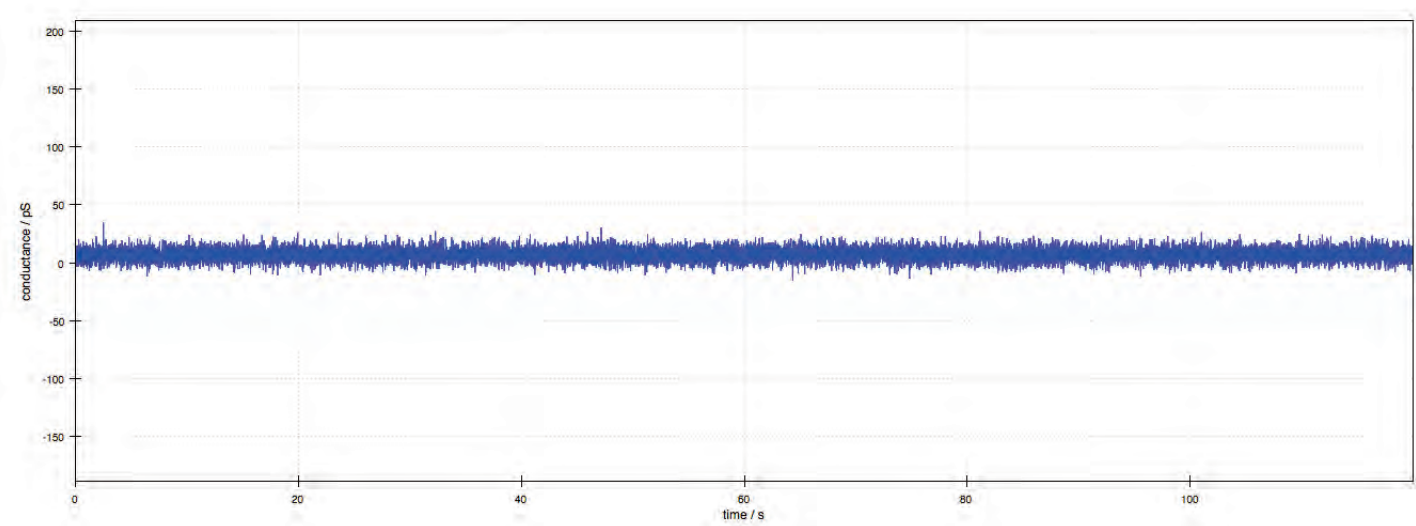


JC-126 bCDMalAm3O1 - JC-V-1 homo 1M CsCl pH 7

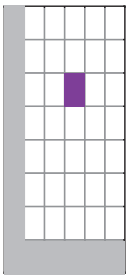
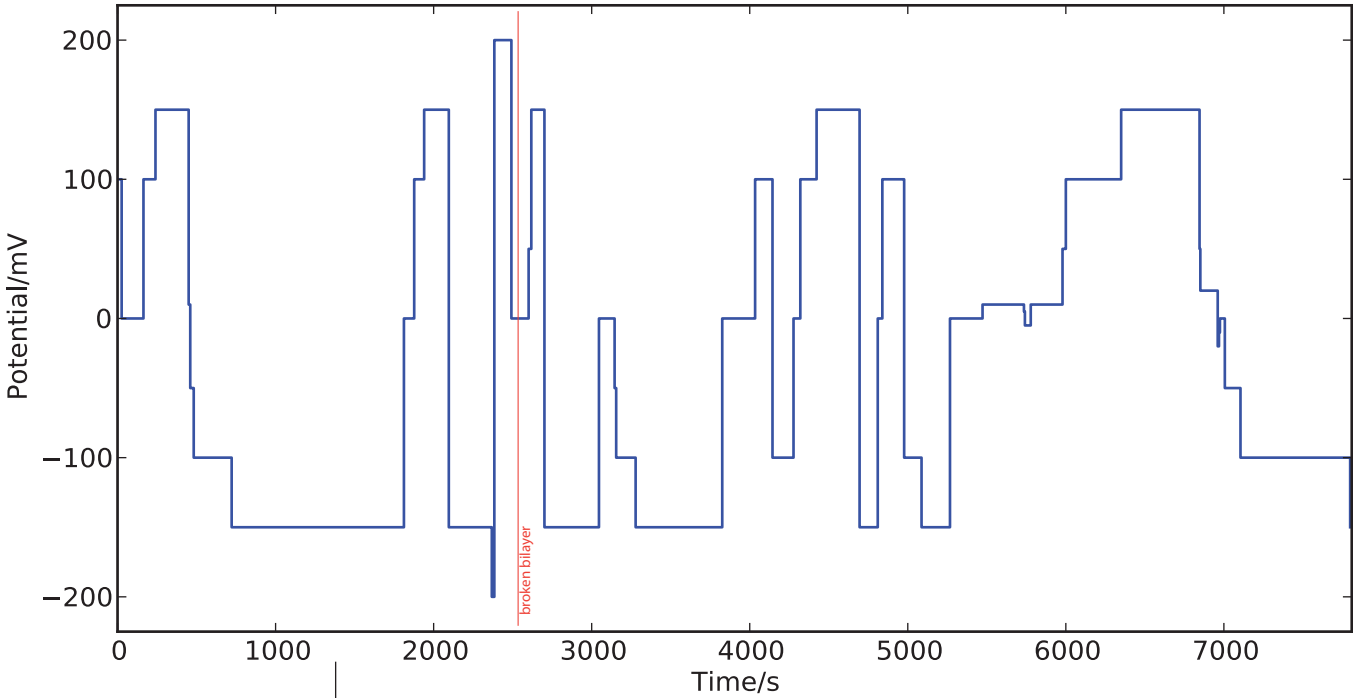


lipid
electrolyte 22/14uM
contact injection
brush transfer

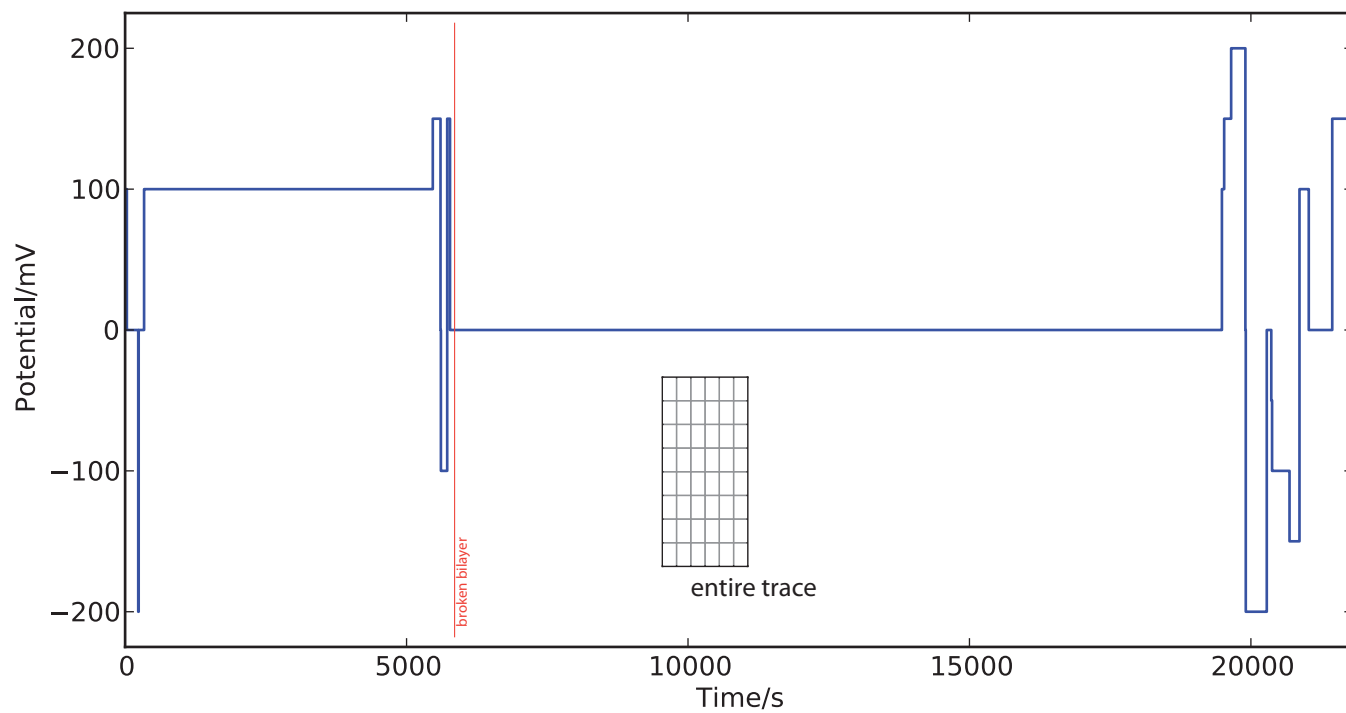




JC-137 JC-V-19 bCD-Mal(Am3O1)2 1percent premix AdNH2 1M CsCl pH 7

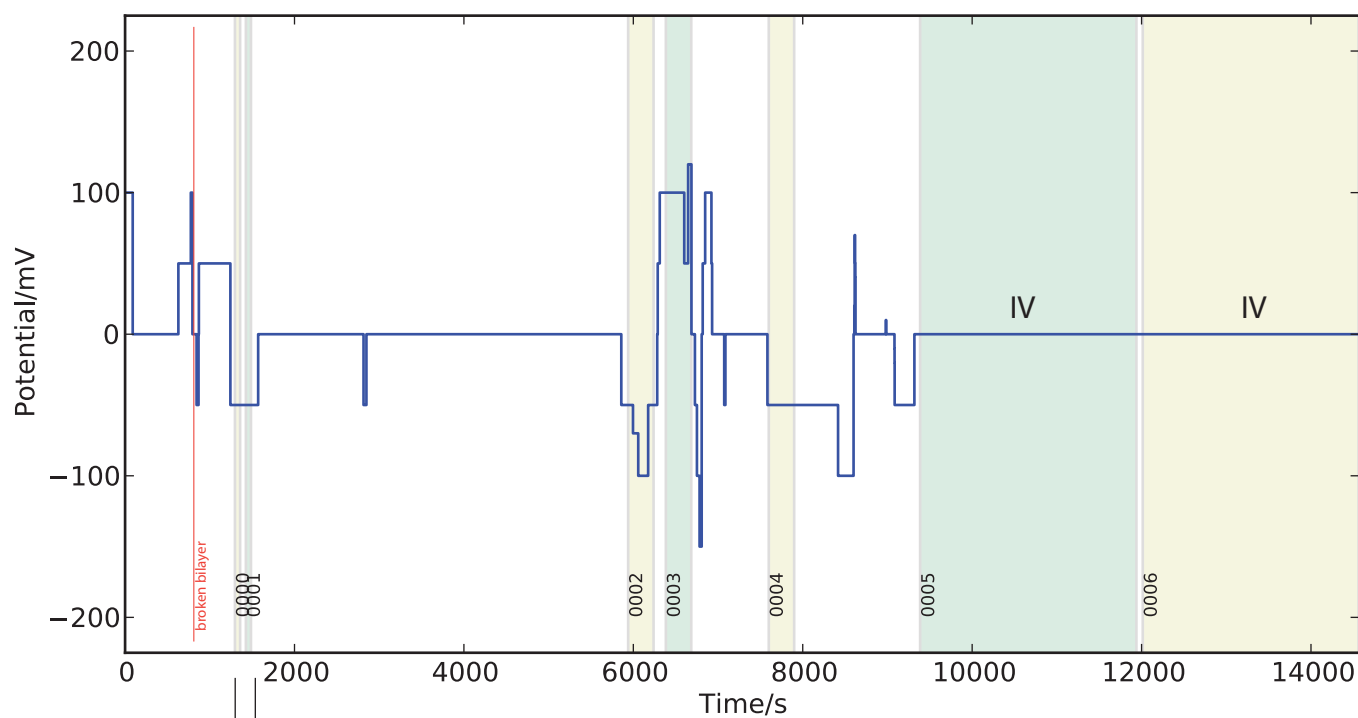


JC-147 bCD-Mal(Am3O1)2 1M NMe4NO3 pH 5 addn AdCOOH

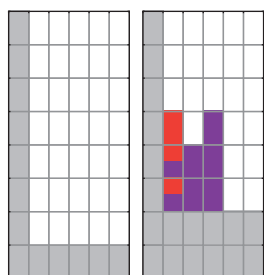


lipid	
electrolyte	
contact injection	
brush transfer	50/30uM total
AdNH2	
lipid	
electrolyte	

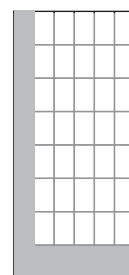
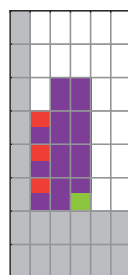
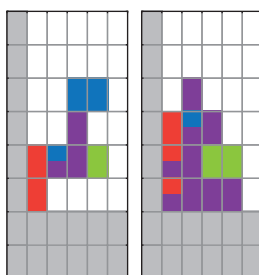
JC-116 bCD2xAm3O2 1M CsCl pH 7



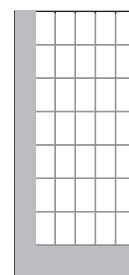
lipid
electrolyte 12.3/7.8uM
contact injection
brush transfer



unfiltered trace filtered trace

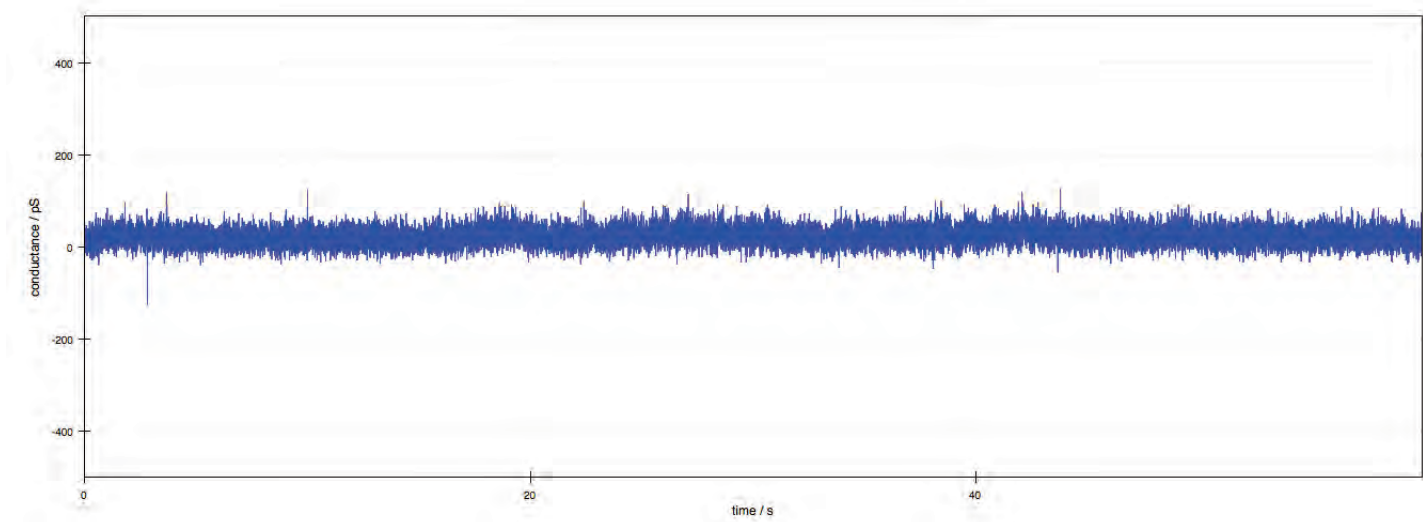


0 -> 120mV,
6mV steps

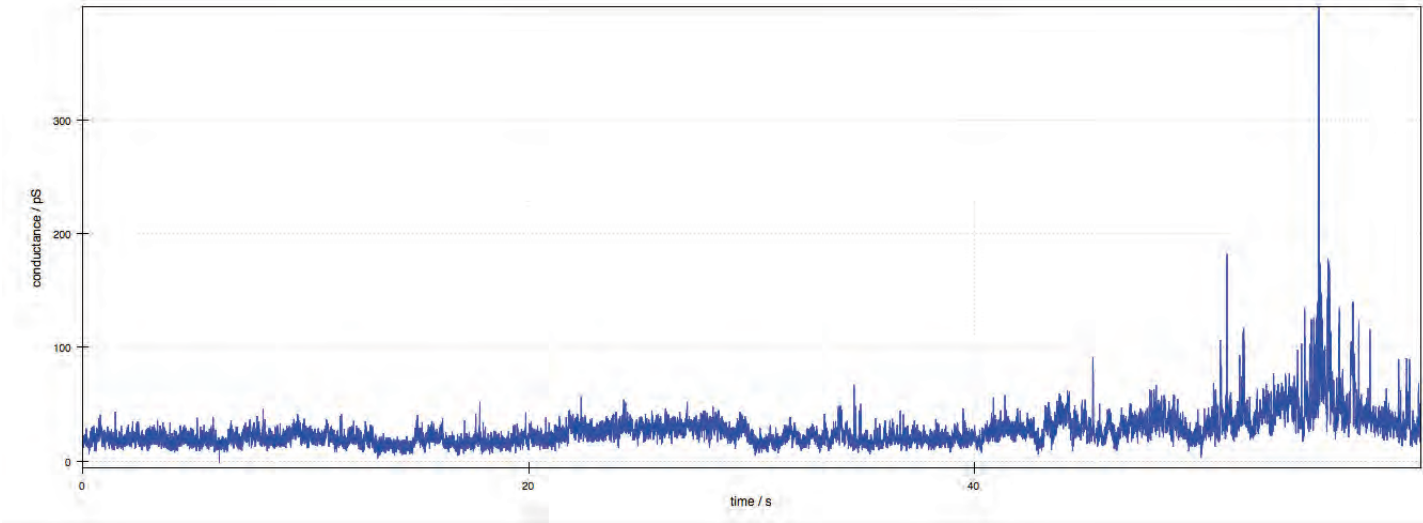


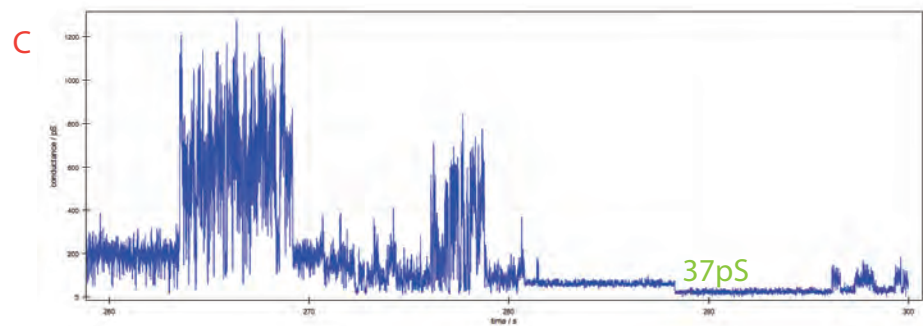
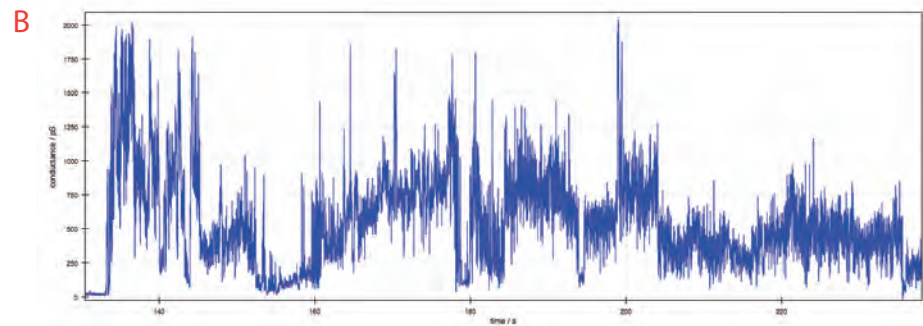
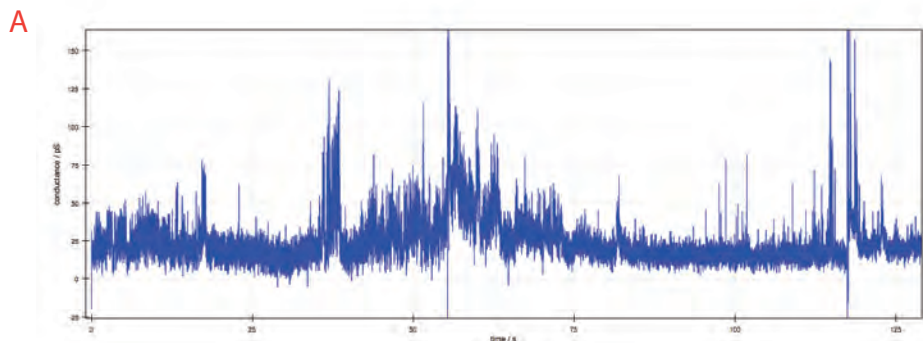
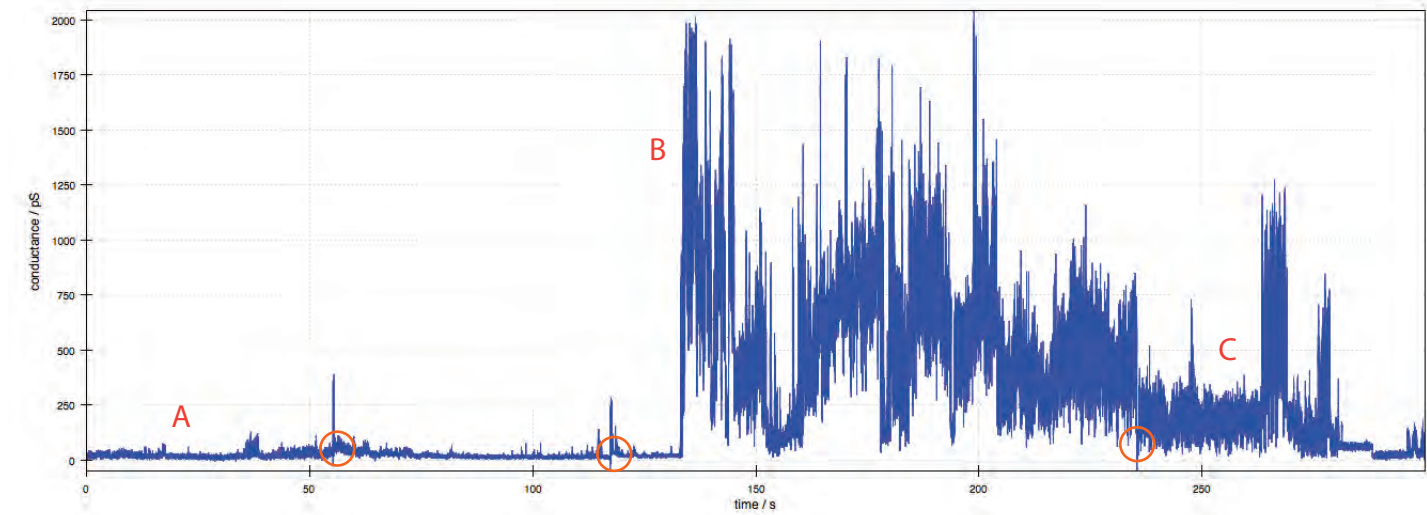
0 -> -200mV,
20mV steps

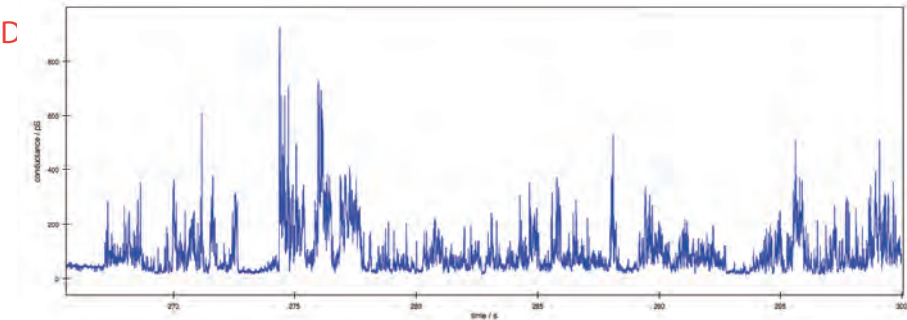
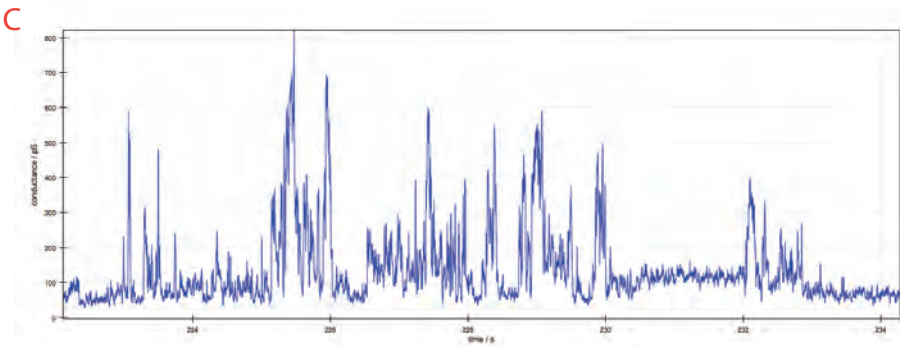
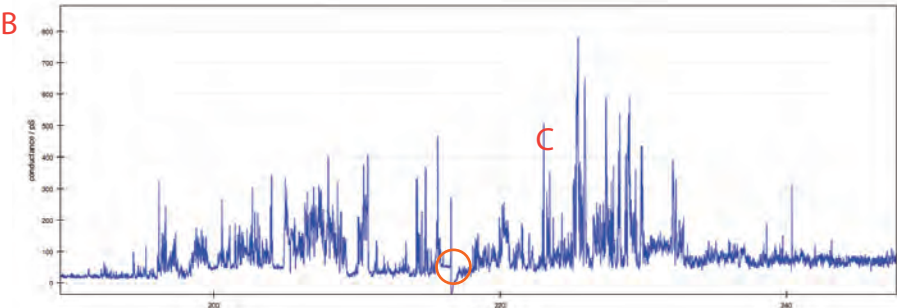
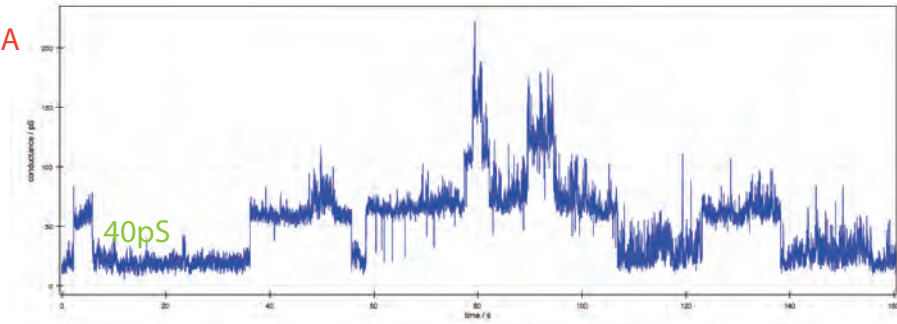
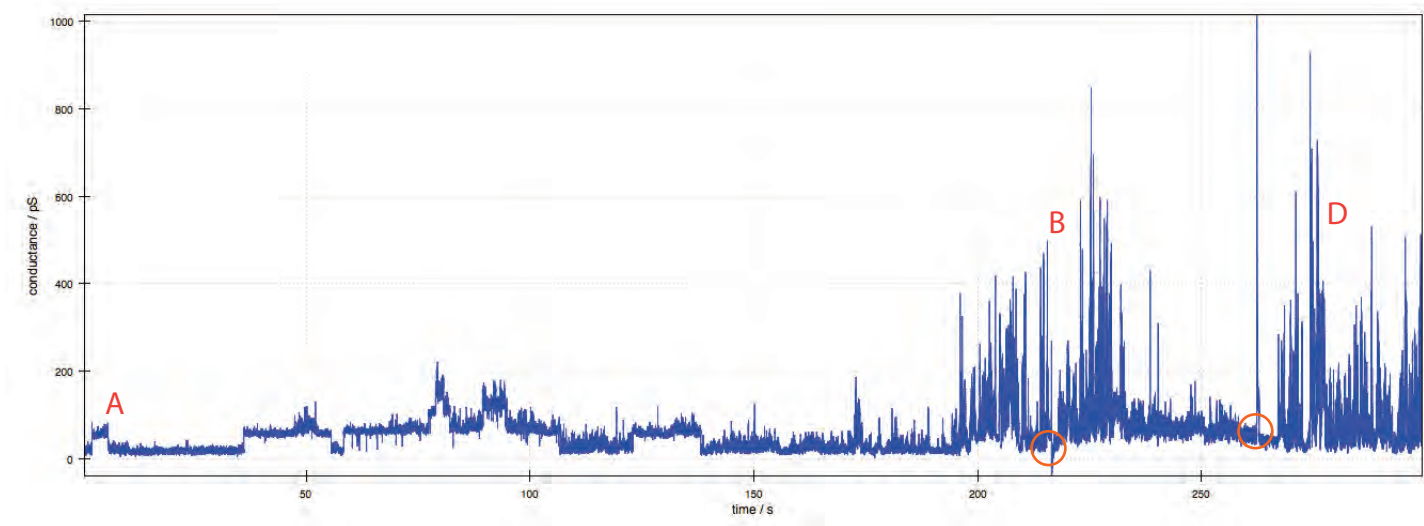
116-0000

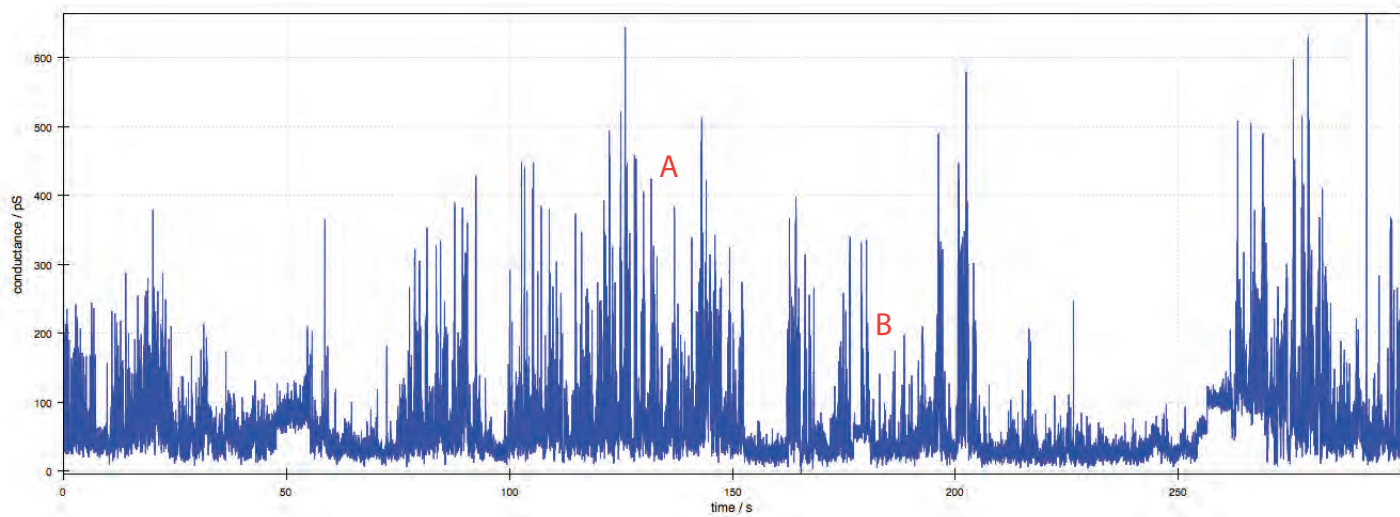


116-0001

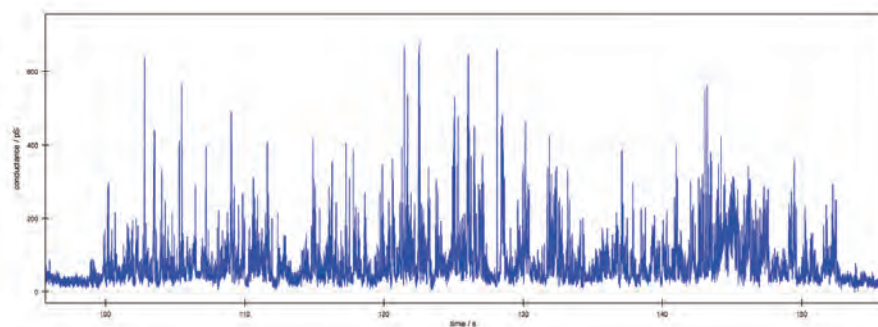




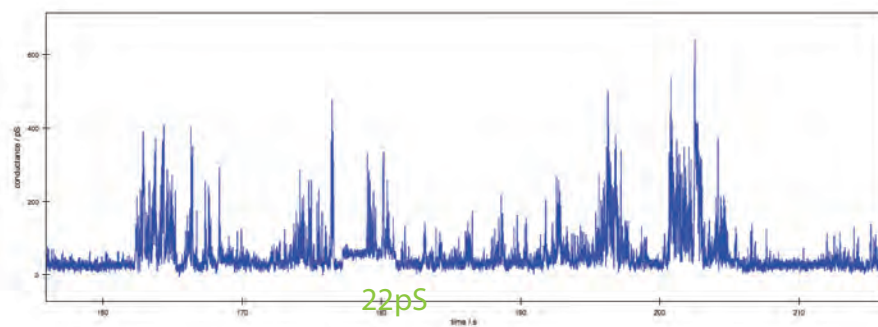




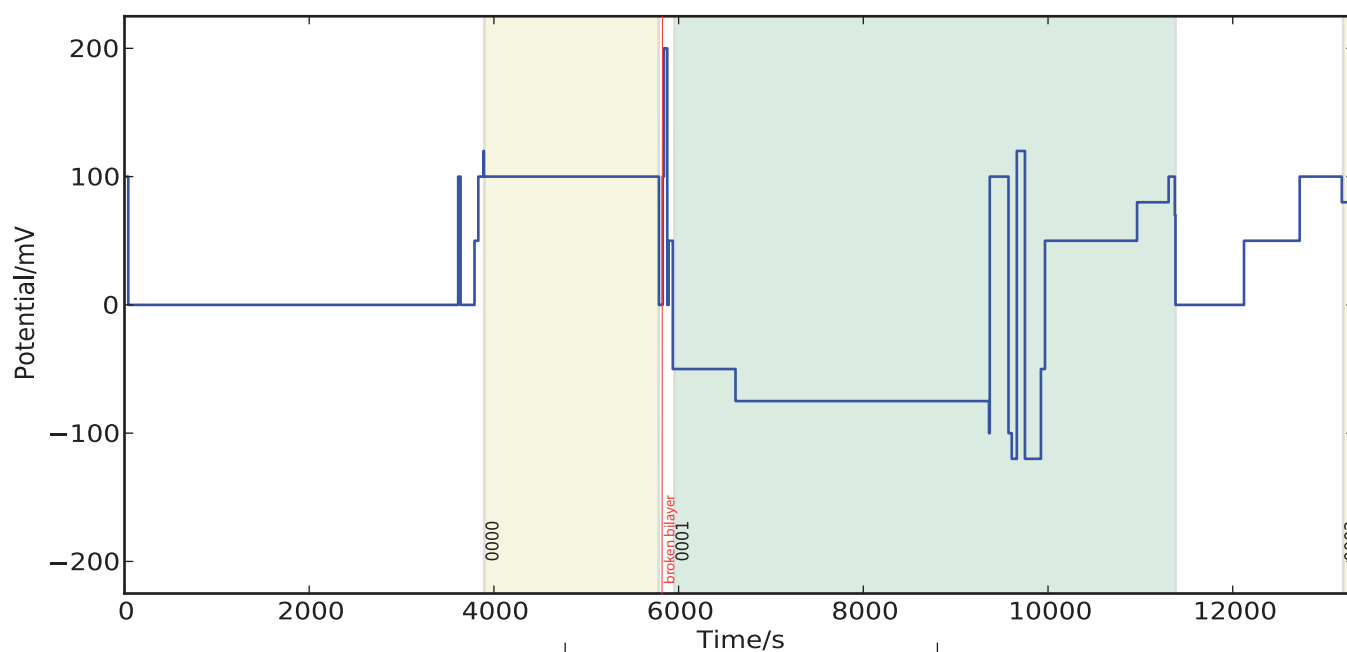
A



B



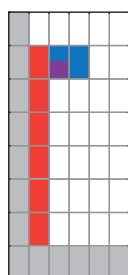
JC-117 bCD2xAm3O2 1M CsCl pH 7 run 2



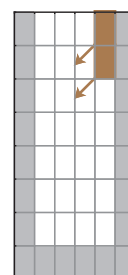
lipid
electrolyte 5.3/3.3uM
contact injection
brush transfer

Adamantyl guest

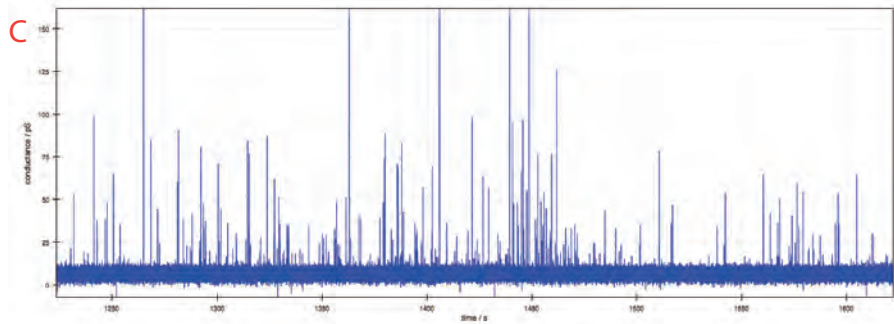
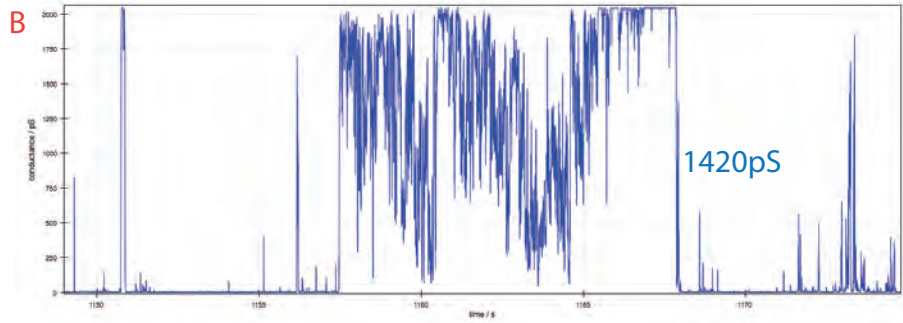
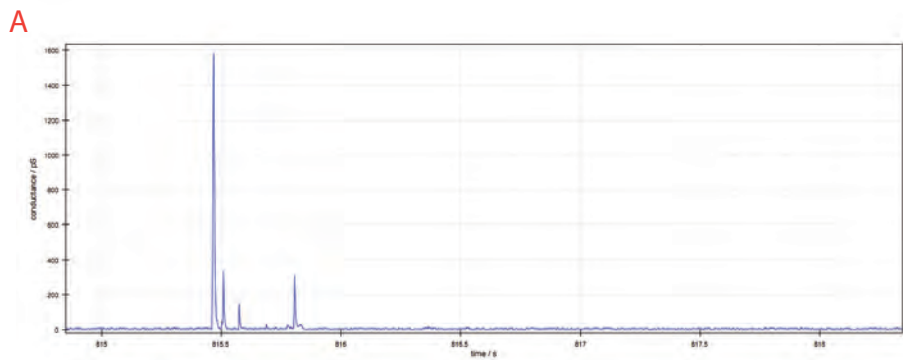
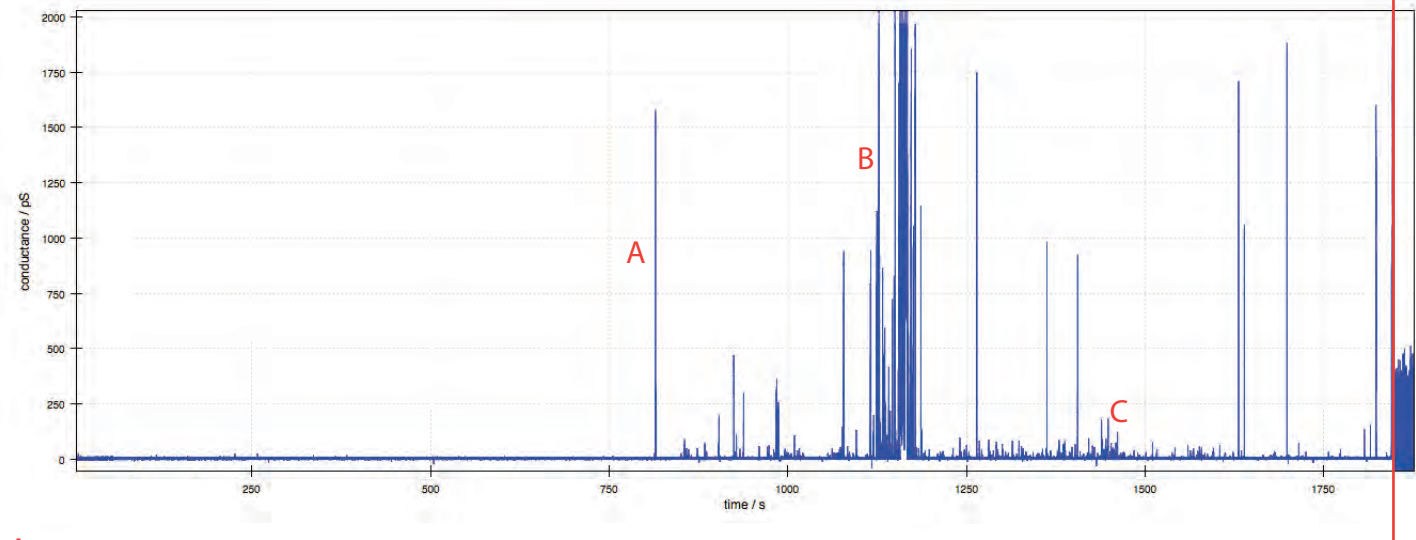
lipid
electrolyte

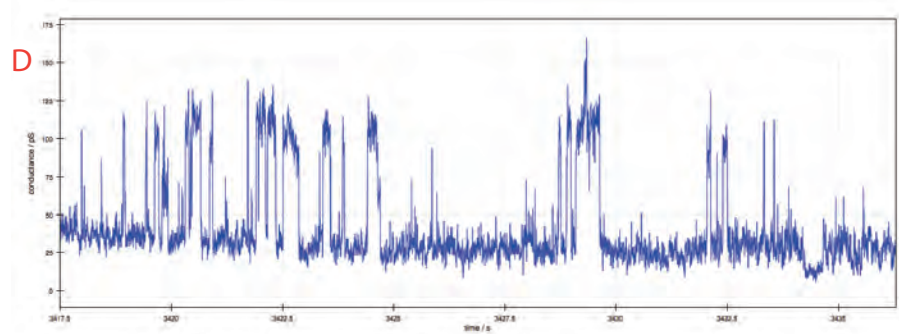
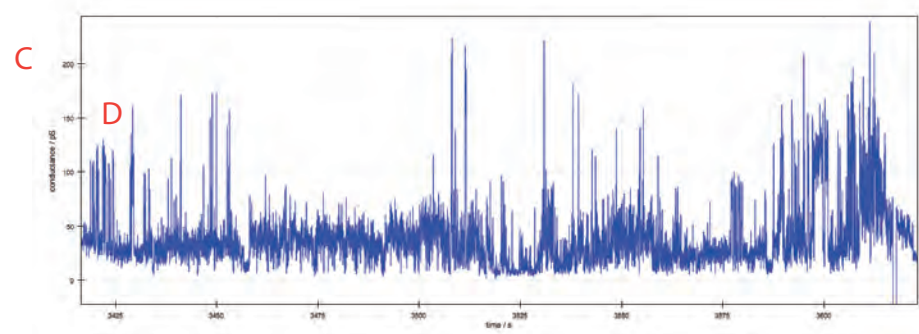
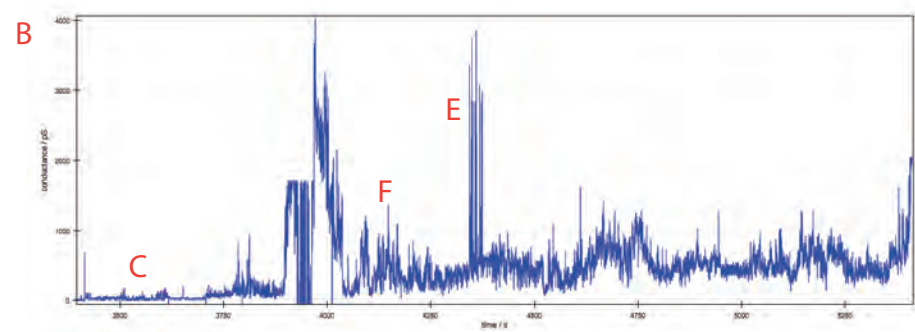
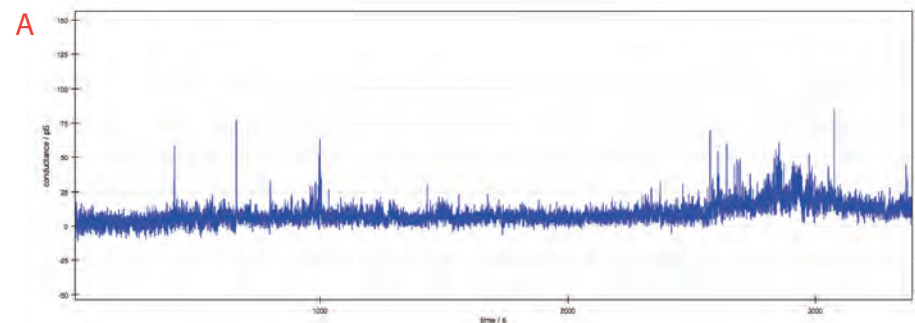
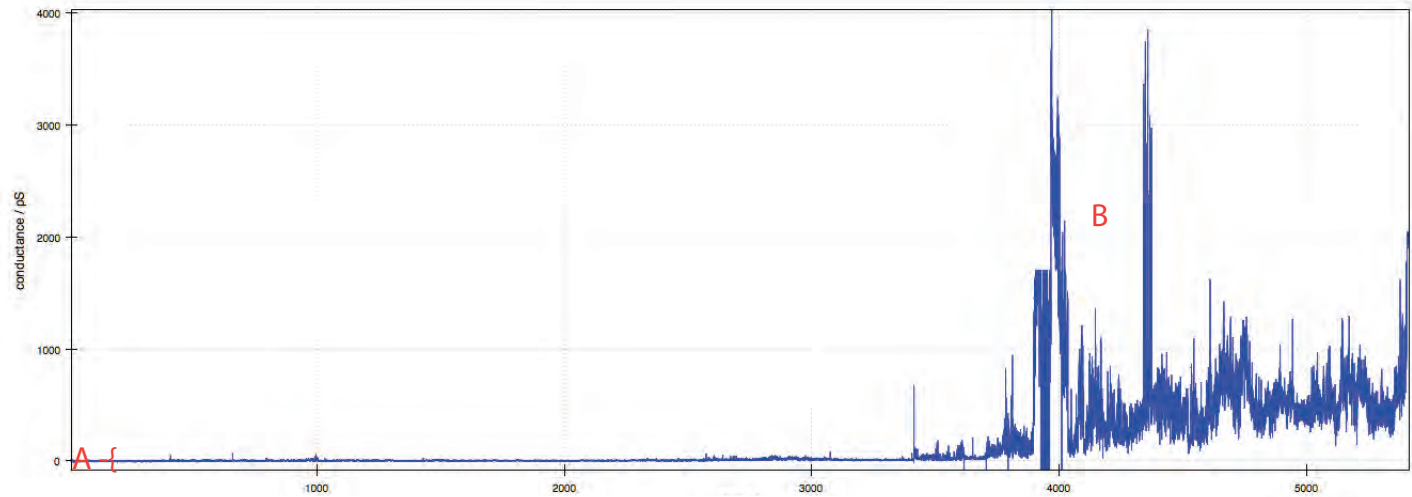


asymmetric?

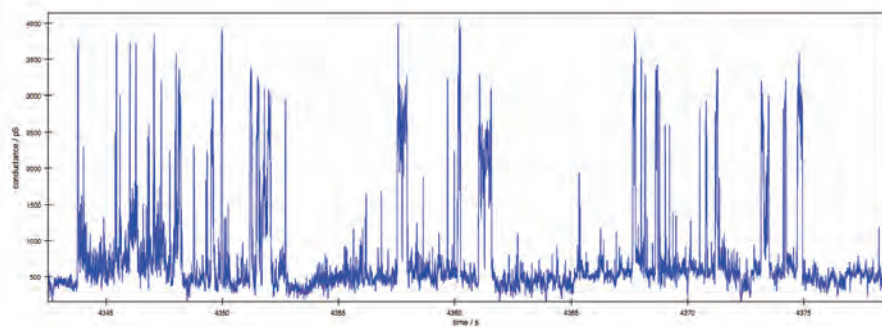


unfiltered trace

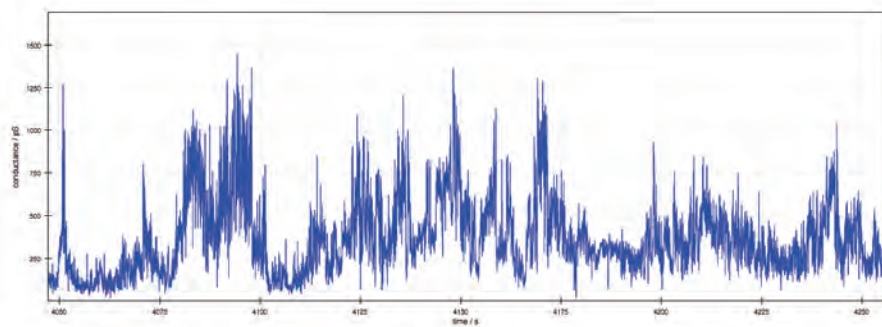




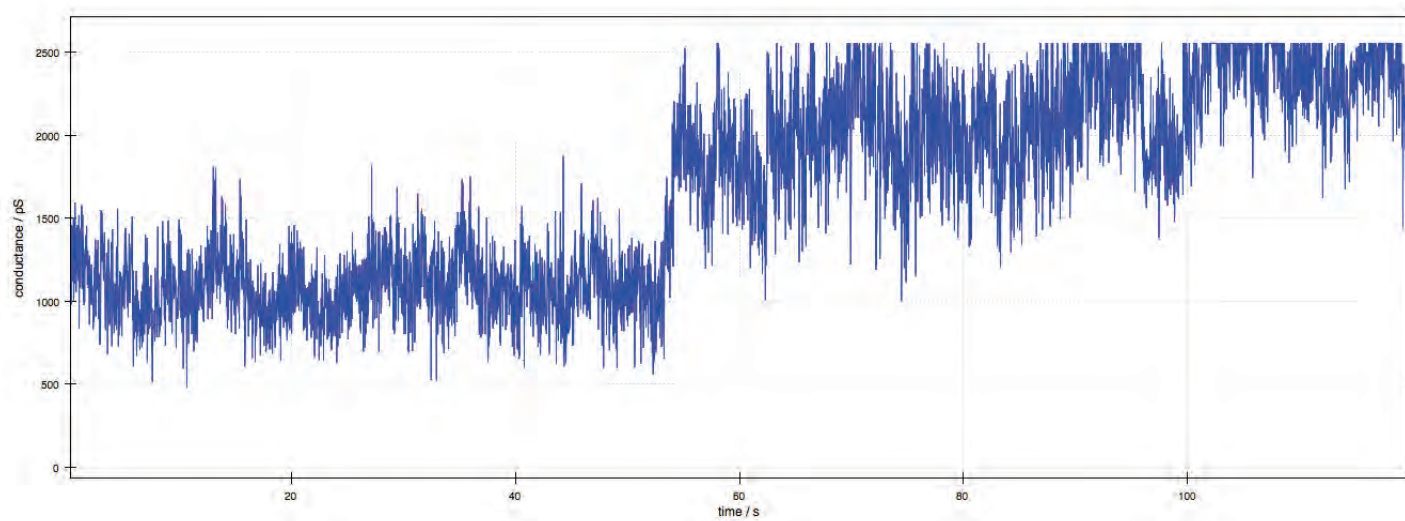
E



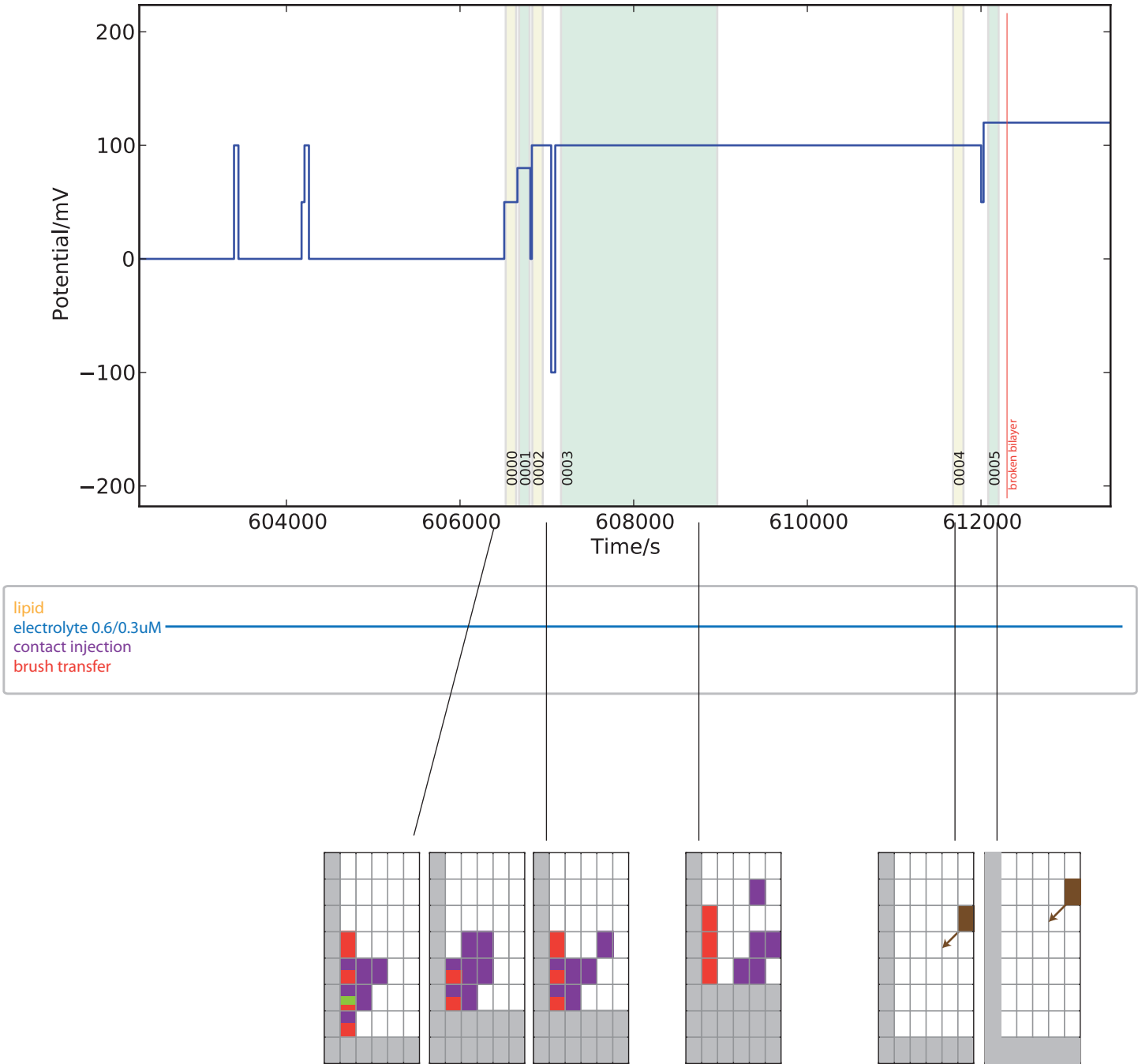
F



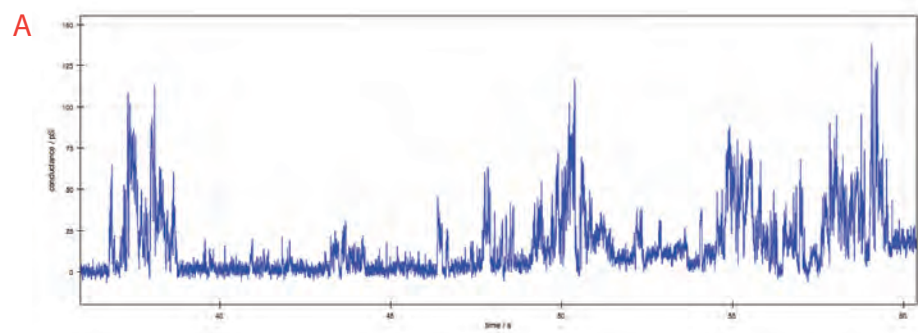
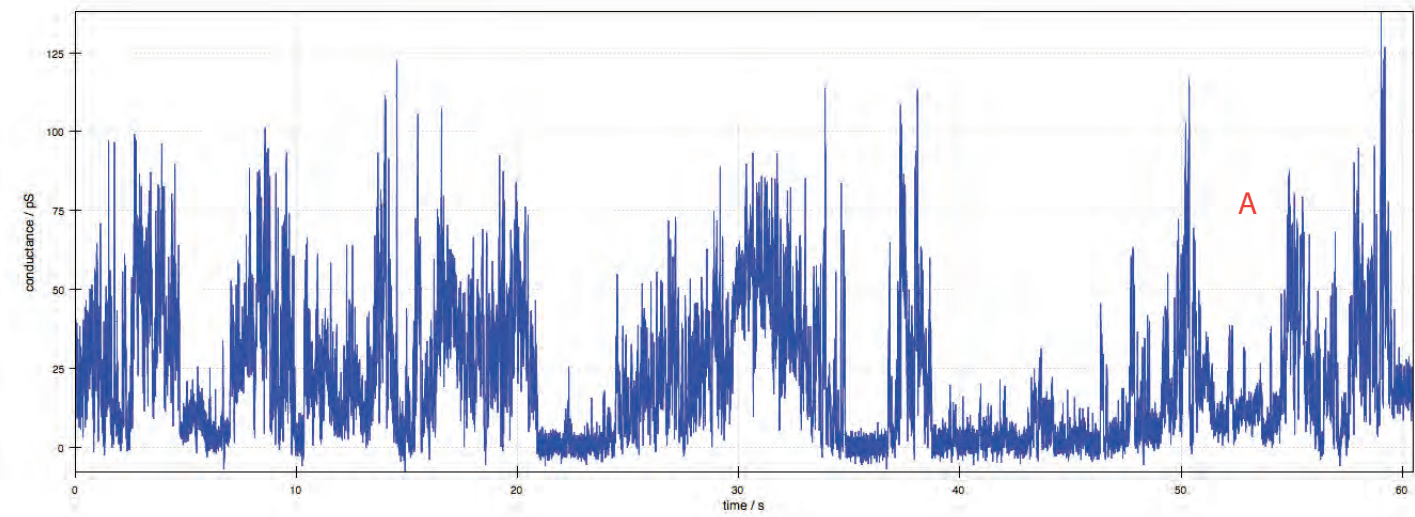
117-0002

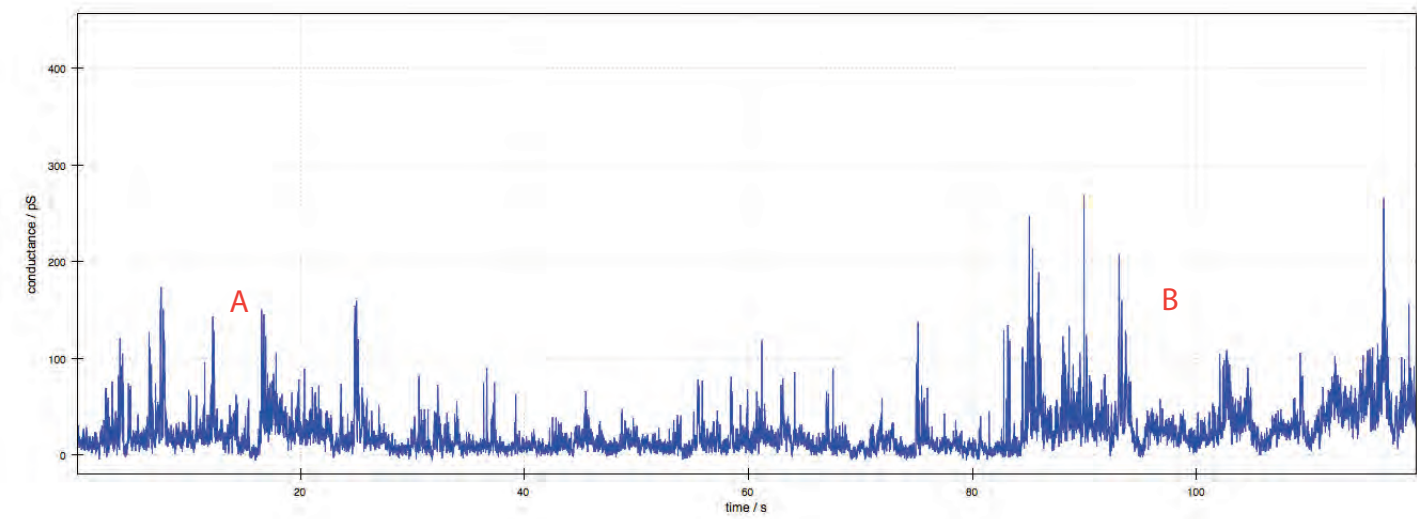


JC-119 bCD2xAm3O2 - impure - 1M CsCl pH 7 run 3

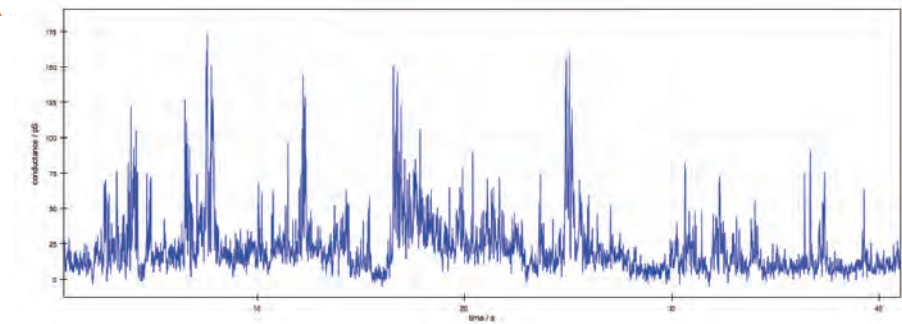


119-0000

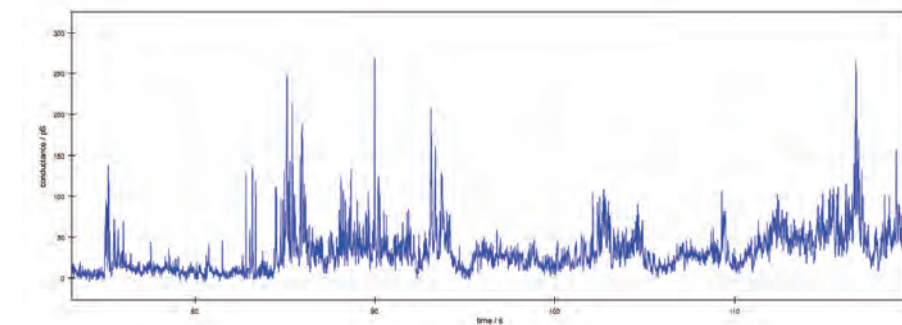




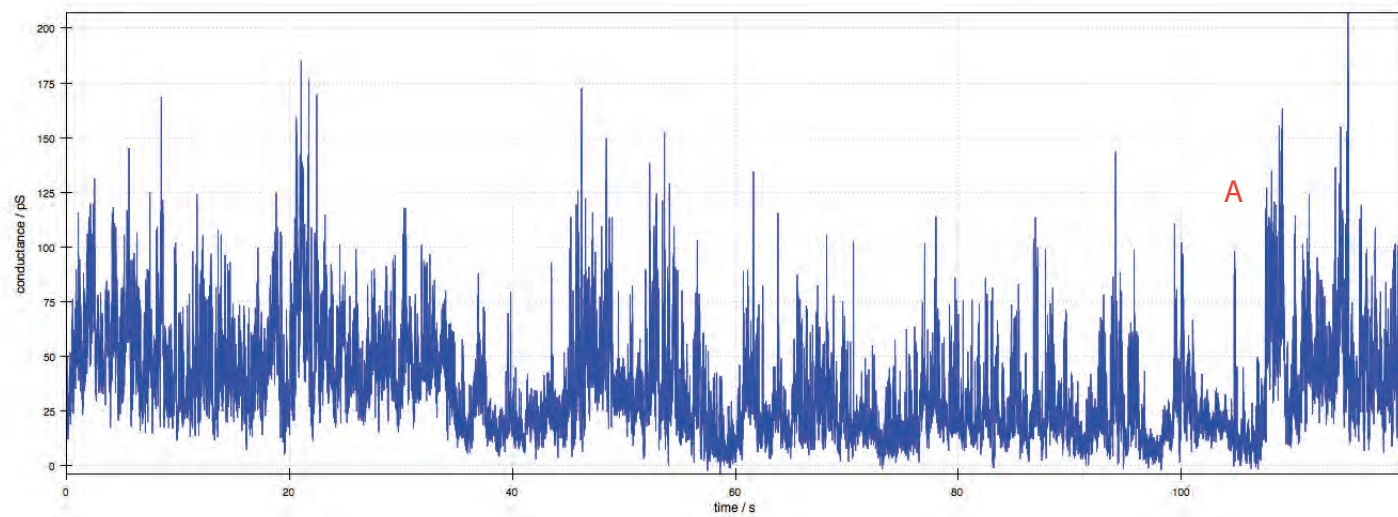
A



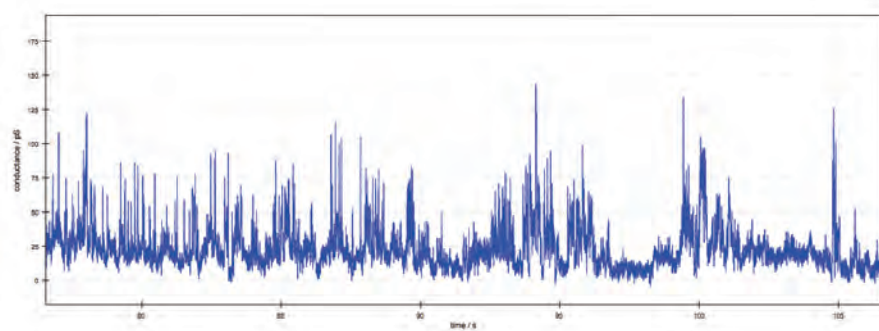
B



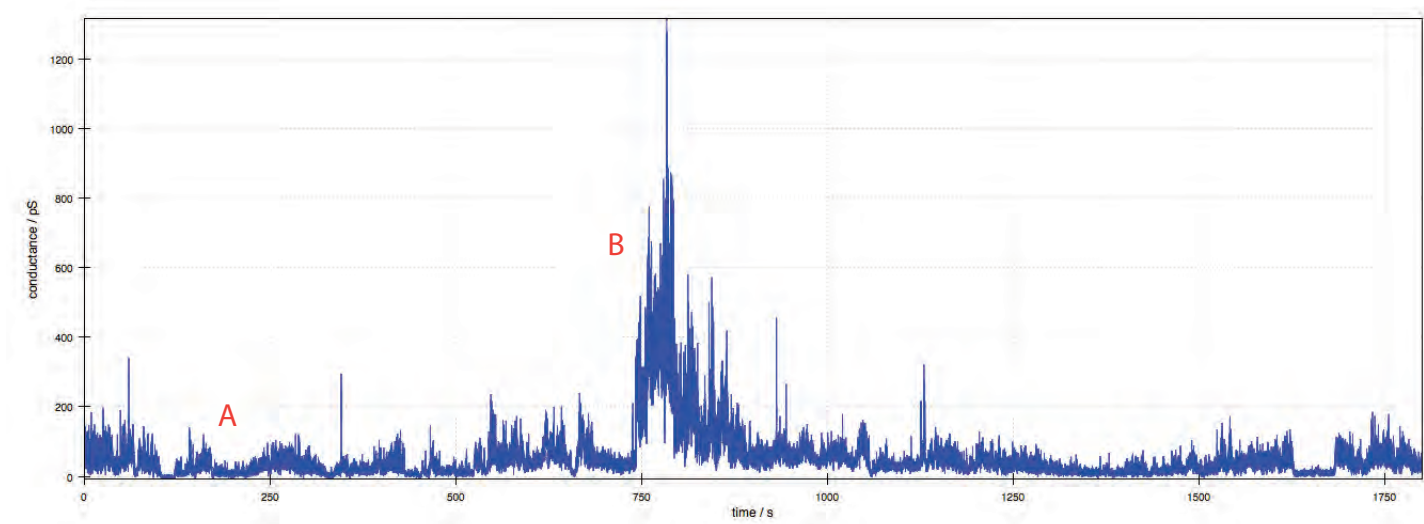
119-0002



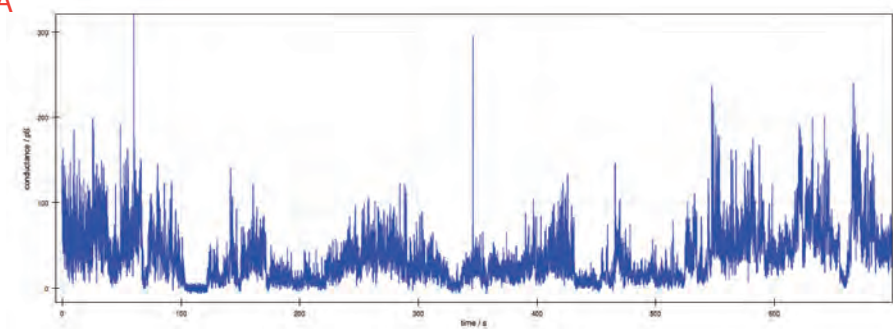
A



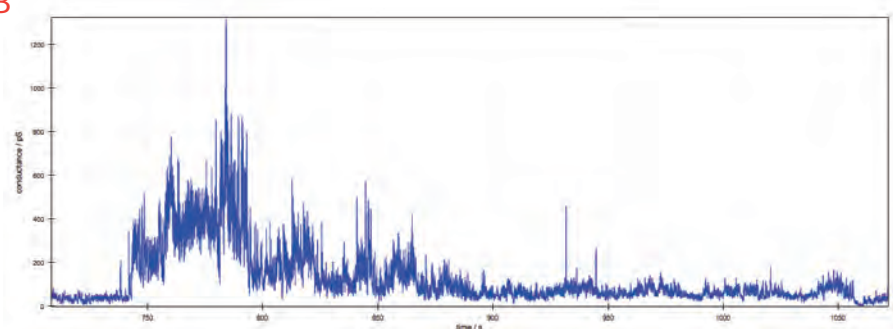
119-0003



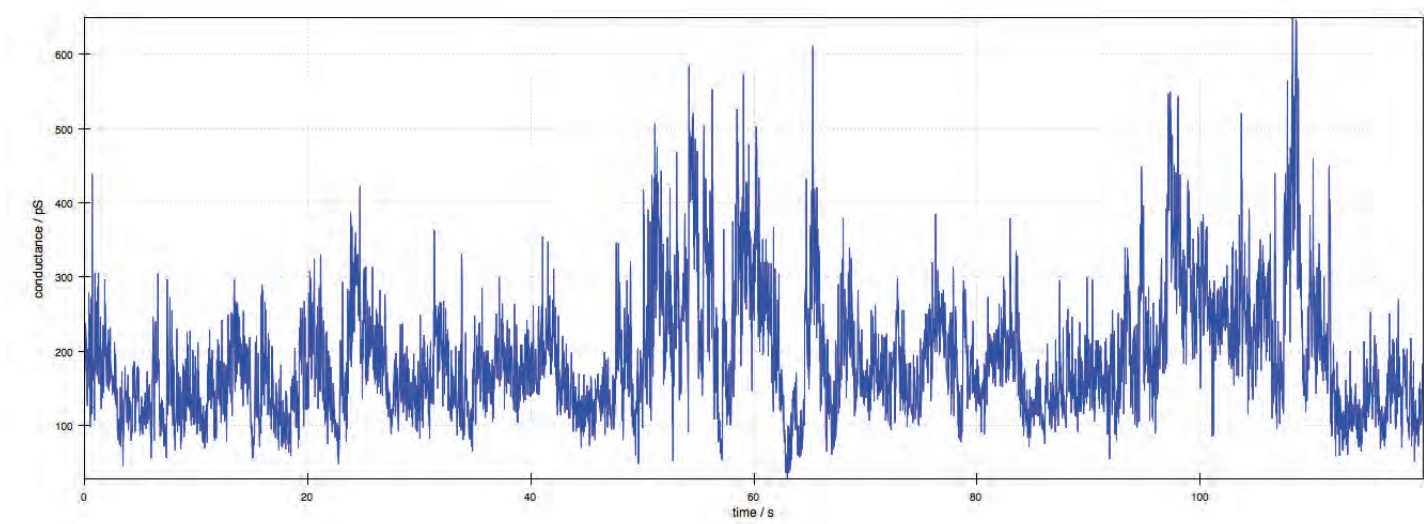
A



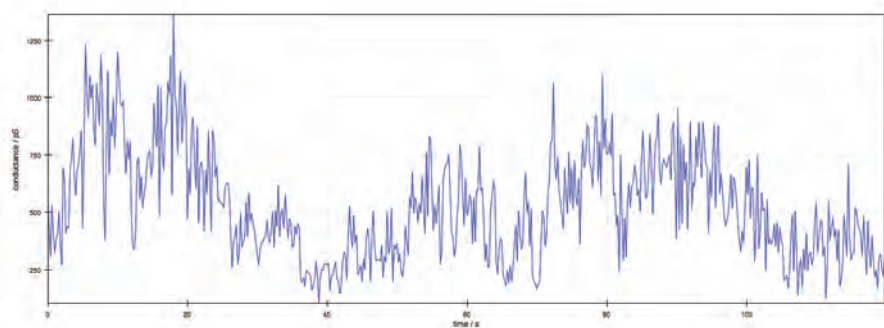
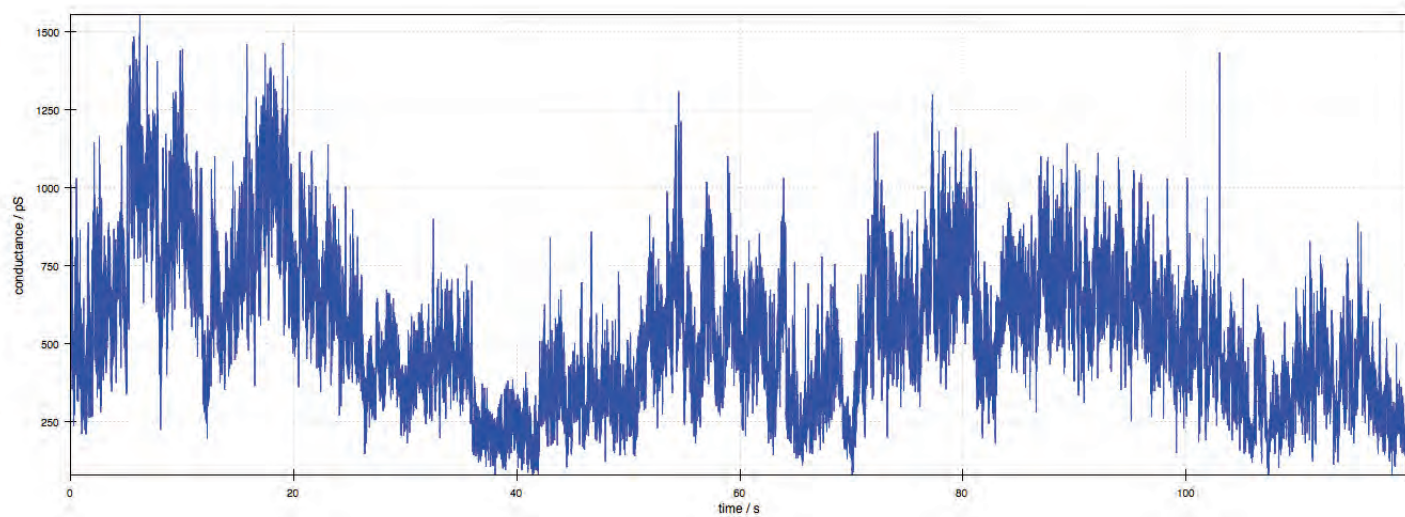
B



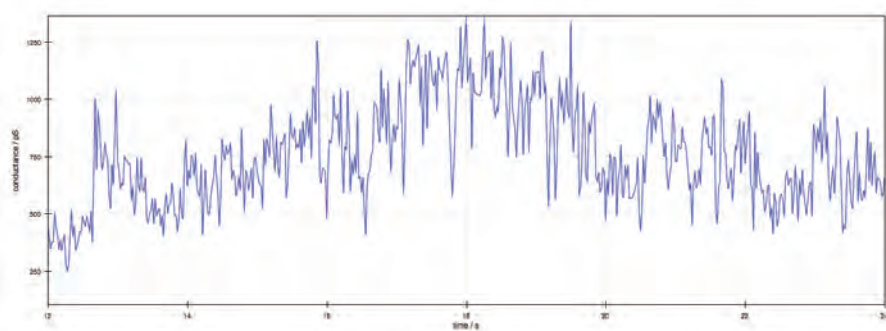
119-0004



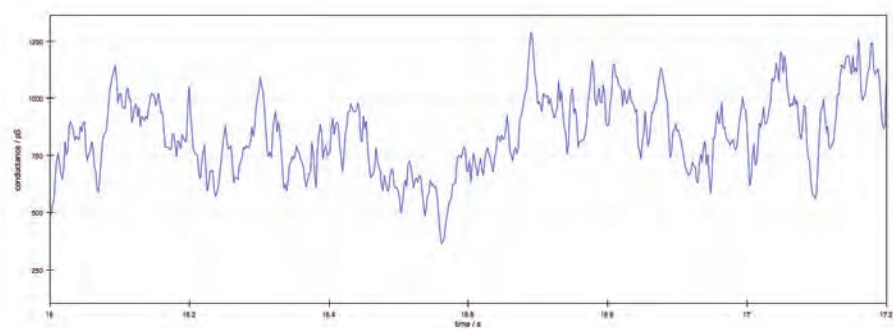
119-0005



1000x decimation



100x decimation



10x decimation